

# Structural and Chemical Features Contributing to Defect Tolerance of Binary Semiconductors

*Or, how can we make **really** cheap solar cell materials?*

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## BLUE WATERS



**Photovoltaics  
Research Laboratory**

# Hybrid perovskites are surprising!

## 1) Astounding efficiency gains

### Best Research-Cell Efficiencies

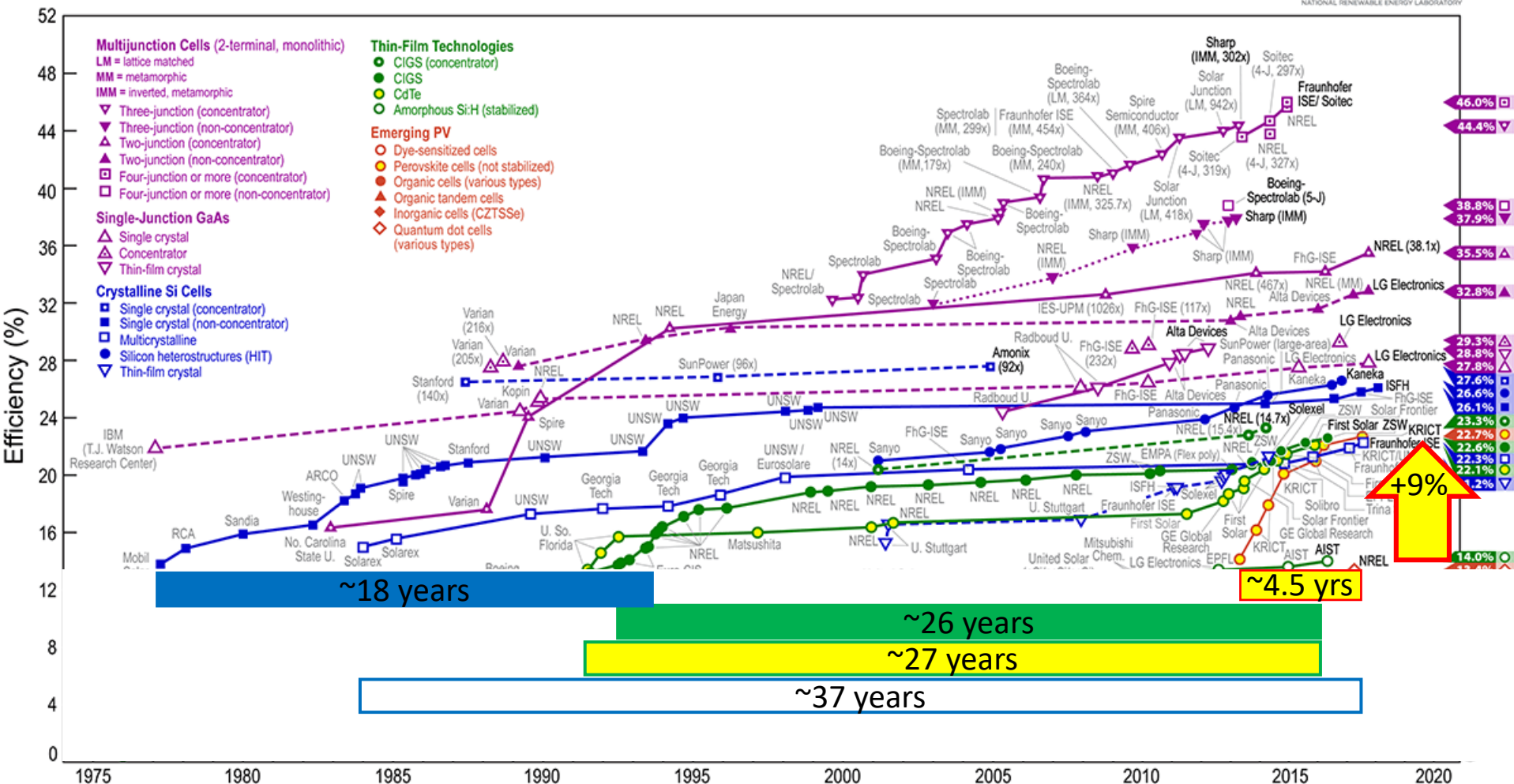


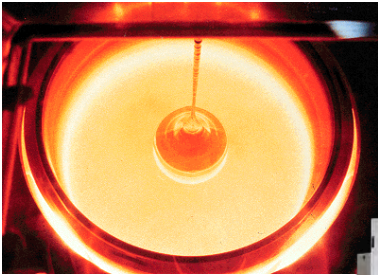
Chart from NREL National Center for Photovoltaics, accessed 2018



# Hybrid perovskites are surprising!

- 1) Astounding efficiency gains
- 2) Unconventional fabrication techniques

Typical high-performance semiconductor manufacturing



High temperature!

High vacuum!



High purity!

So how are they so good?!?\*

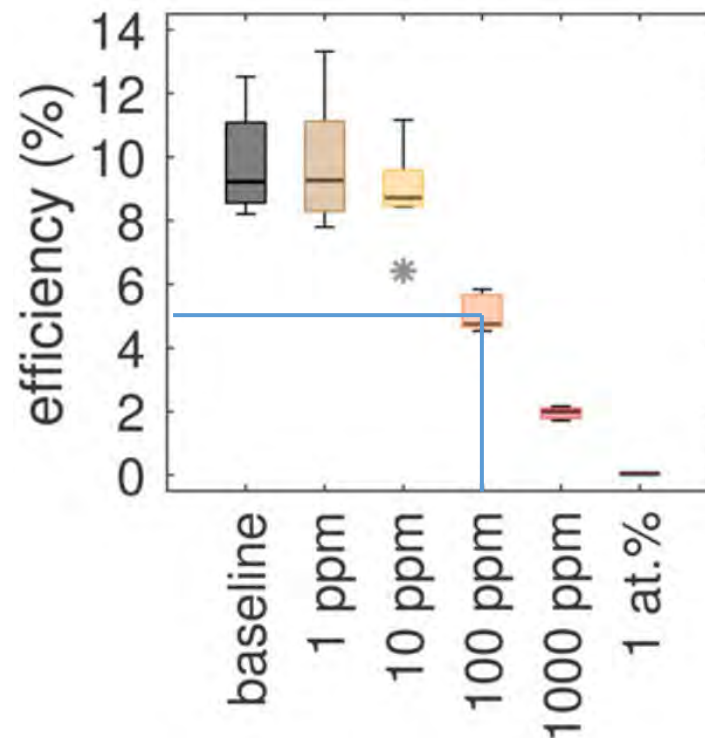
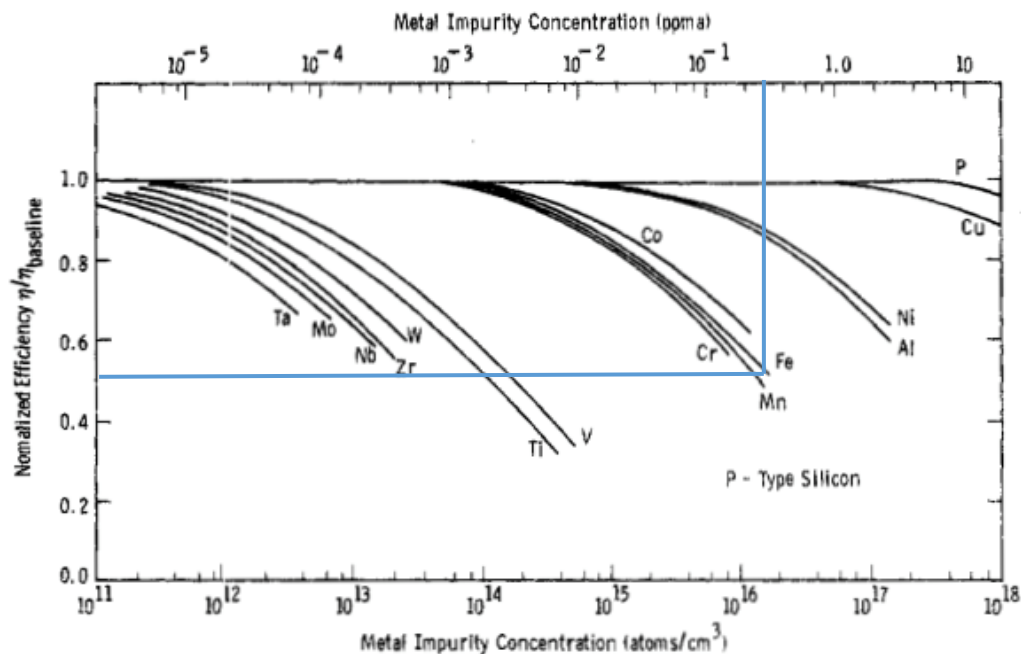
Hybrid perovskites manufacturing:

Ambient temperature and pressure, precursors and processing done in a normal glovebox in a regular lab!



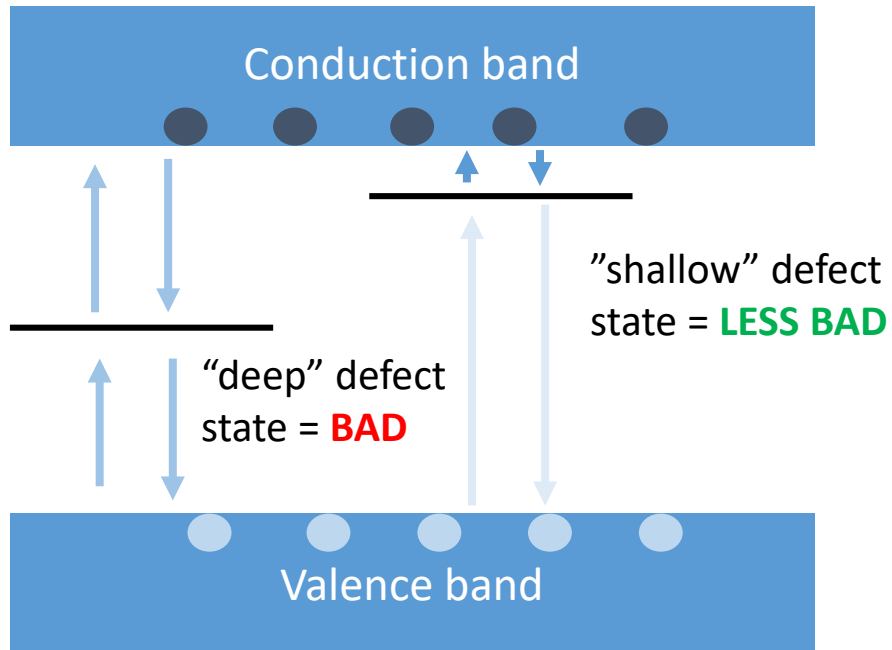
\* Other than the toxicity and lack of stability...

# Defect Tolerance!



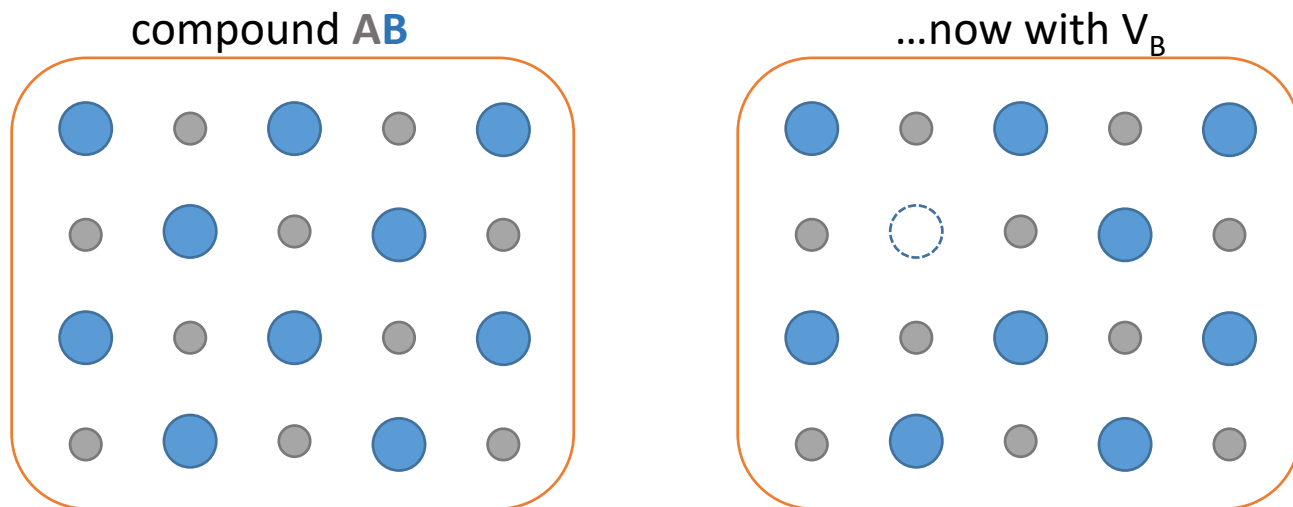
50% relative efficiency loss at over 100 TIMES the concentration of iron!

# Why are shallow defects good?



$$U_{\text{SRH}} \propto e^{|E_t - E_i|}$$

# DFT Defect Calculations



$$\Delta H_{D,q}(E_F) \approx [E_{D,q} - E_{\text{host}}] + \underline{qE_F} + \underbrace{\sum_{\alpha} n_{\alpha} \mu_{\alpha}} + \underline{E_{\text{corr}}}$$

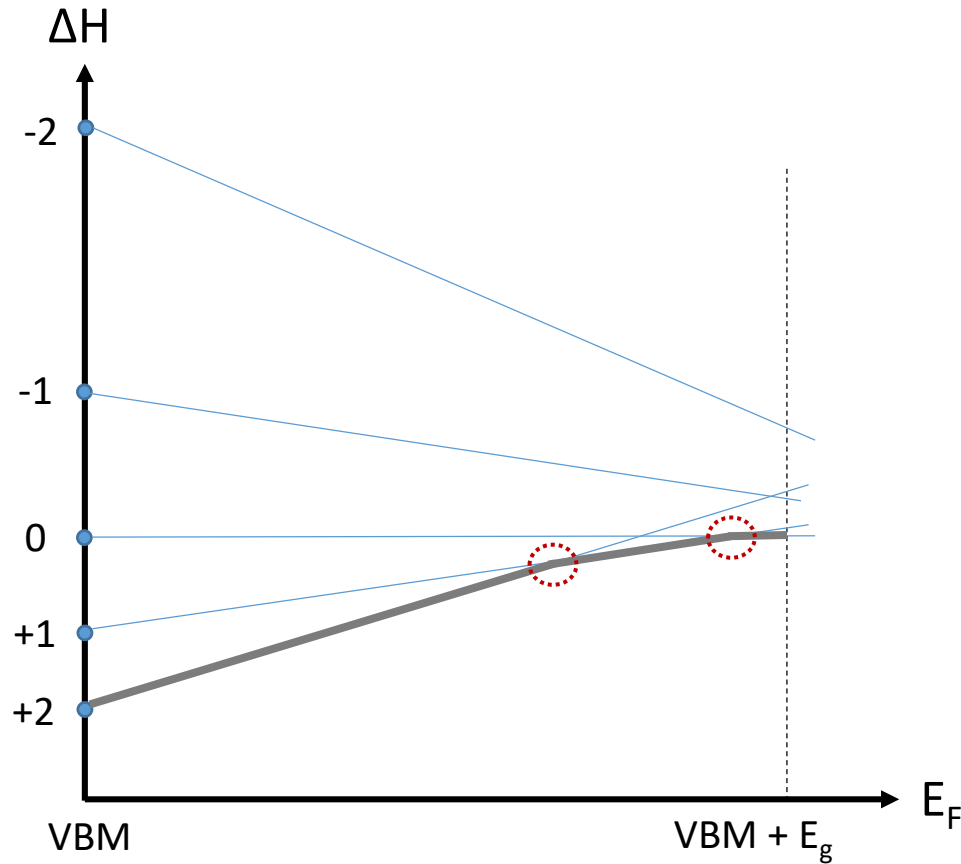
Electrochemical potential  
(electrons)

Chemical potential (for  
atoms added/removed)  
(here  $1 \times \mu_B$ )

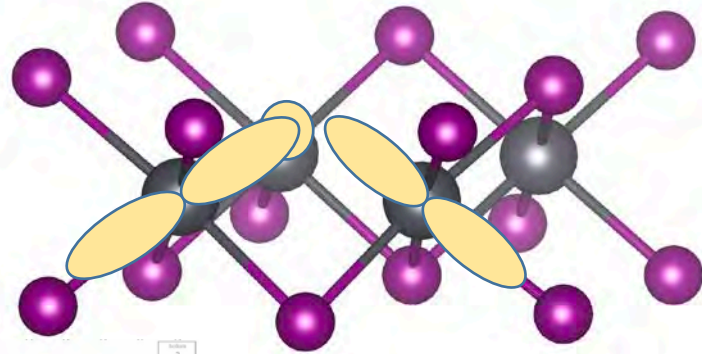
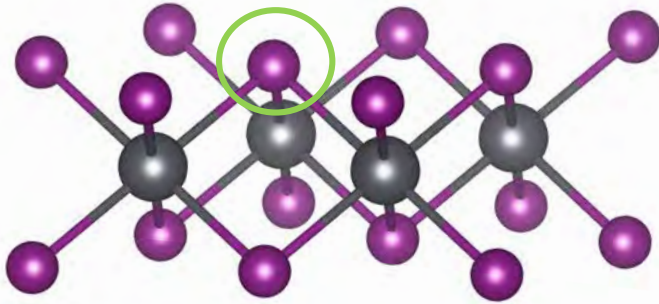
Compensate for  
supercell size effects

# Defect Energy Diagrams

$$\Delta H_{D,q}(E_F) \approx [E_{D,q} - E_{\text{host}}] + qE_F + \sum_{\alpha} n_{\alpha} \mu_{\alpha} + E_{\text{corr}}$$



# How does defect tolerance happen?



1	H	2	He																																
3	Li	4	Be																																
5	B	6	C	7	N	8	O	9	F	10	Ne																								
11	Na	12	Mg																																
13	Al	14	Si	15	P	16	S	17	Cl	18	Ar																								
19	K	20	Ca	21	Sc	22	Ti	23	V	24	Cr	25	Mn	26	Fe	27	Co	28	Ni	29	Cu	30	Zn	31	Ga	32	Ge	33	As	34	Se	35	Br	36	Kr
37	Rb	38	Sr	39	Y	40	Zr	41	Nb	42	Mo	43	Tc	44	Ru	45	Rh	46	Pd	47	Ag	48	Cd	49	In	50	Sn	51	Sb	52	Te	53	I	54	Xe
55	Cs	56	Ba	57-70	Lu	71	Hf	72	Ta	73	W	74	Re	75	Os	76	Ir	77	Pt	78	Au	79	Hg	80	Tl	81	Pb	82	Bi	83	Po	84	At	85	Rn
87	Fr	88	Ra	**	**	89	Lr	103	Rf	104	Db	105	Sg	106	Bh	107	Hs	108	Mt	109	Uun	110	Uuu	111	Uub	112	Uuq								

\* Lanthanide series

57	La	58	Ce	59	Pr	60	Nd	61	Pm	62	Sm	63	Eu	64	Gd	65	Tb	66	Dy	67	Ho	68	Er	69	Tm	70	Yb
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\*\* Actinide series

89	Ac	90	Th	91	Pa	92	U	93	Np	94	Pu	95	Am	96	Cm	97	Bk	98	Cf	99	Es	100	Fm	101	Mn	102	Nv
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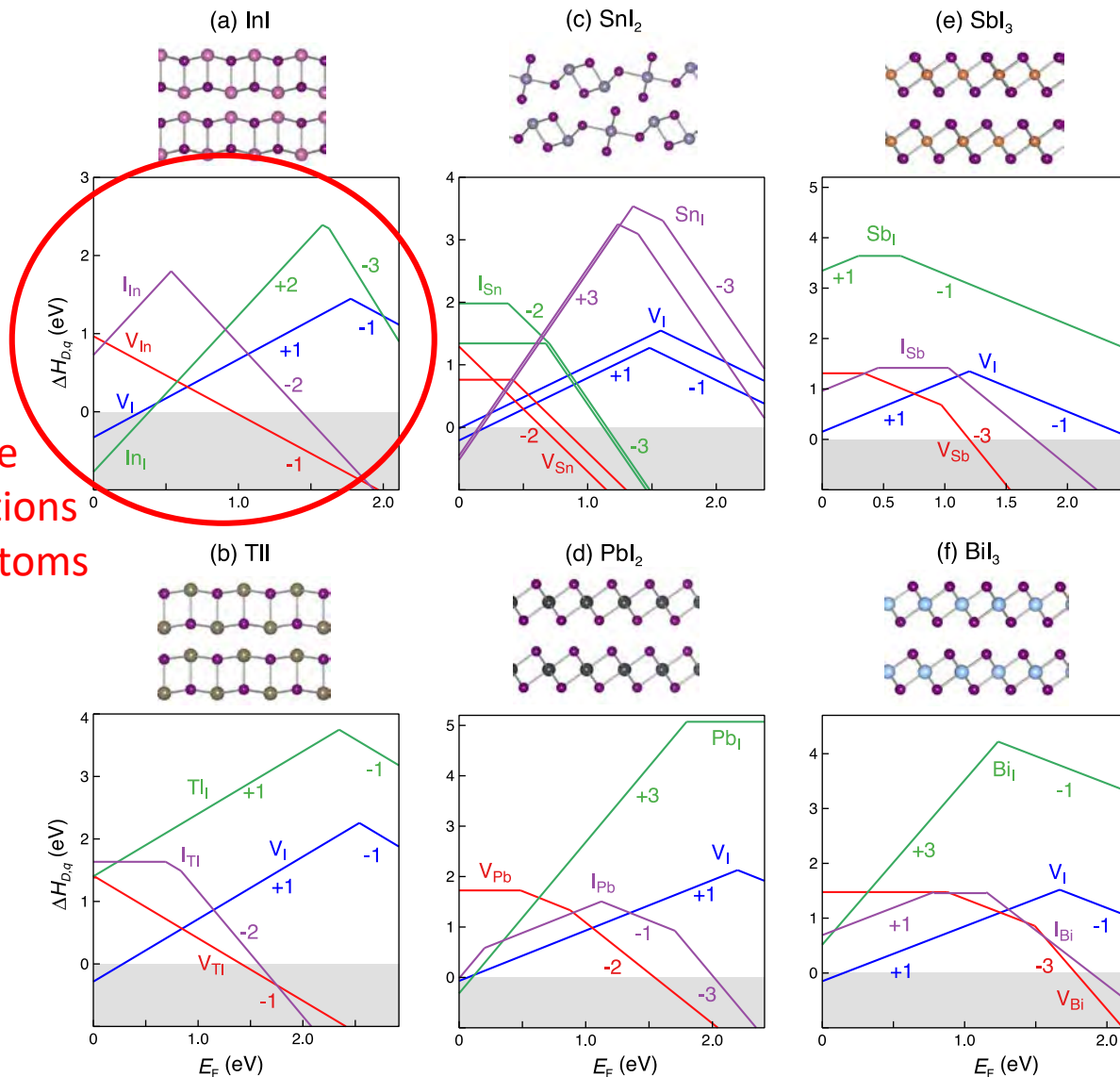
<b>In</b> <b>+1,+3</b>	<b>Sn</b> <b>+2,+4</b>	<b>Sb</b> <b>+3,+5,</b> <b>-3,-2</b>
<b>Tl</b> <b>+1,+3</b>	<b>Pb</b> <b>+2,+4</b>	<b>Bi</b> <b>+3,+5,</b> <b>-3,-2</b>



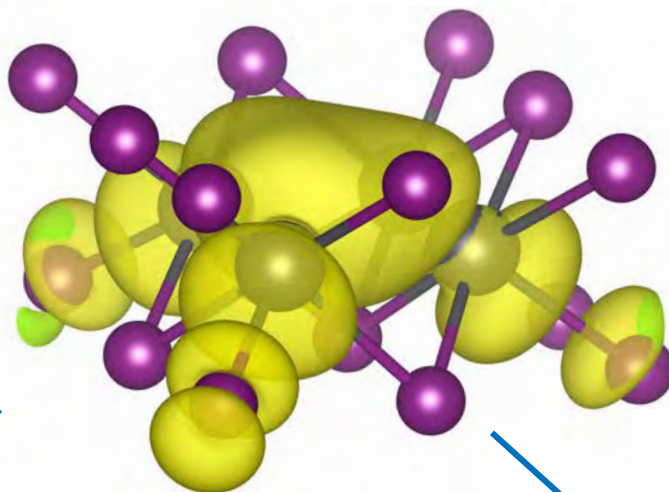
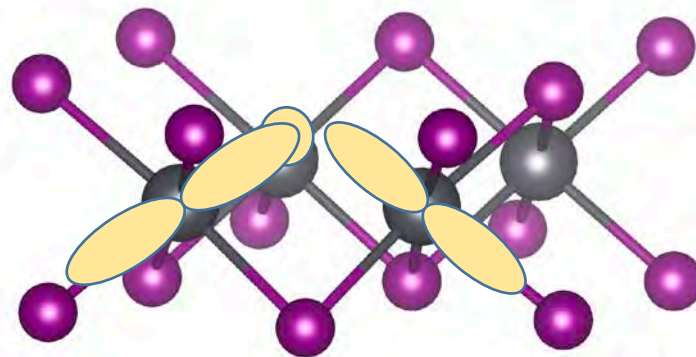
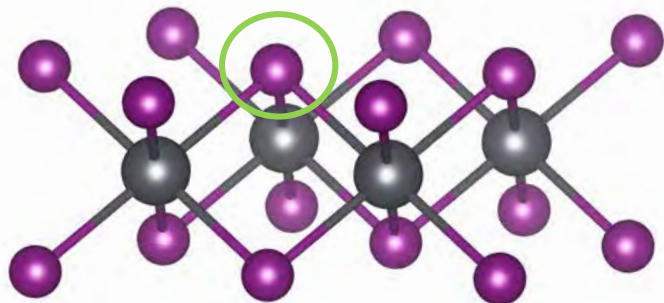


# But they don't have shallow defects!

~30 separate  
DFT calculations  
on  $\mathcal{O}(100)$  atoms



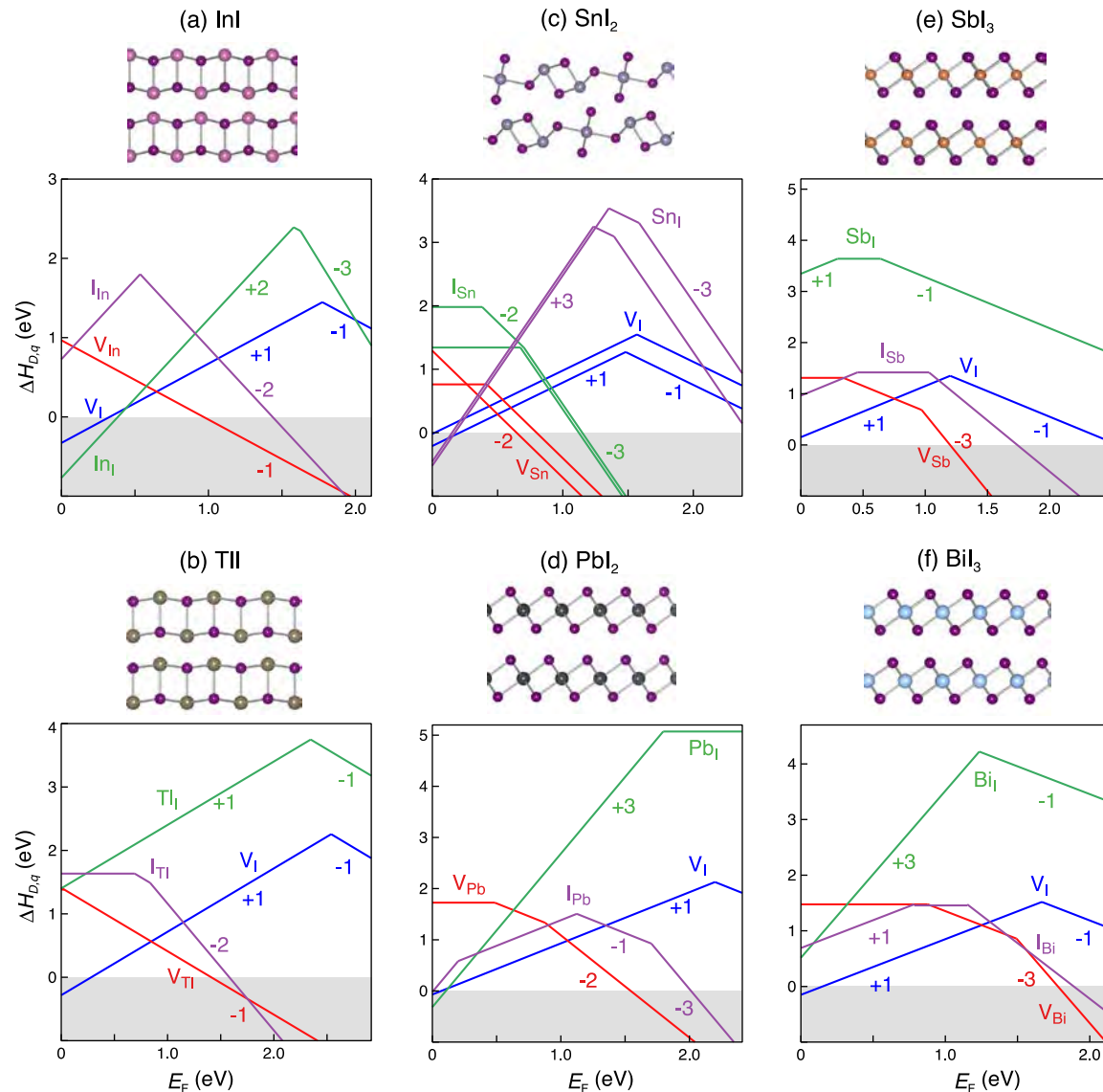
# Updating our understanding



change the **structure**

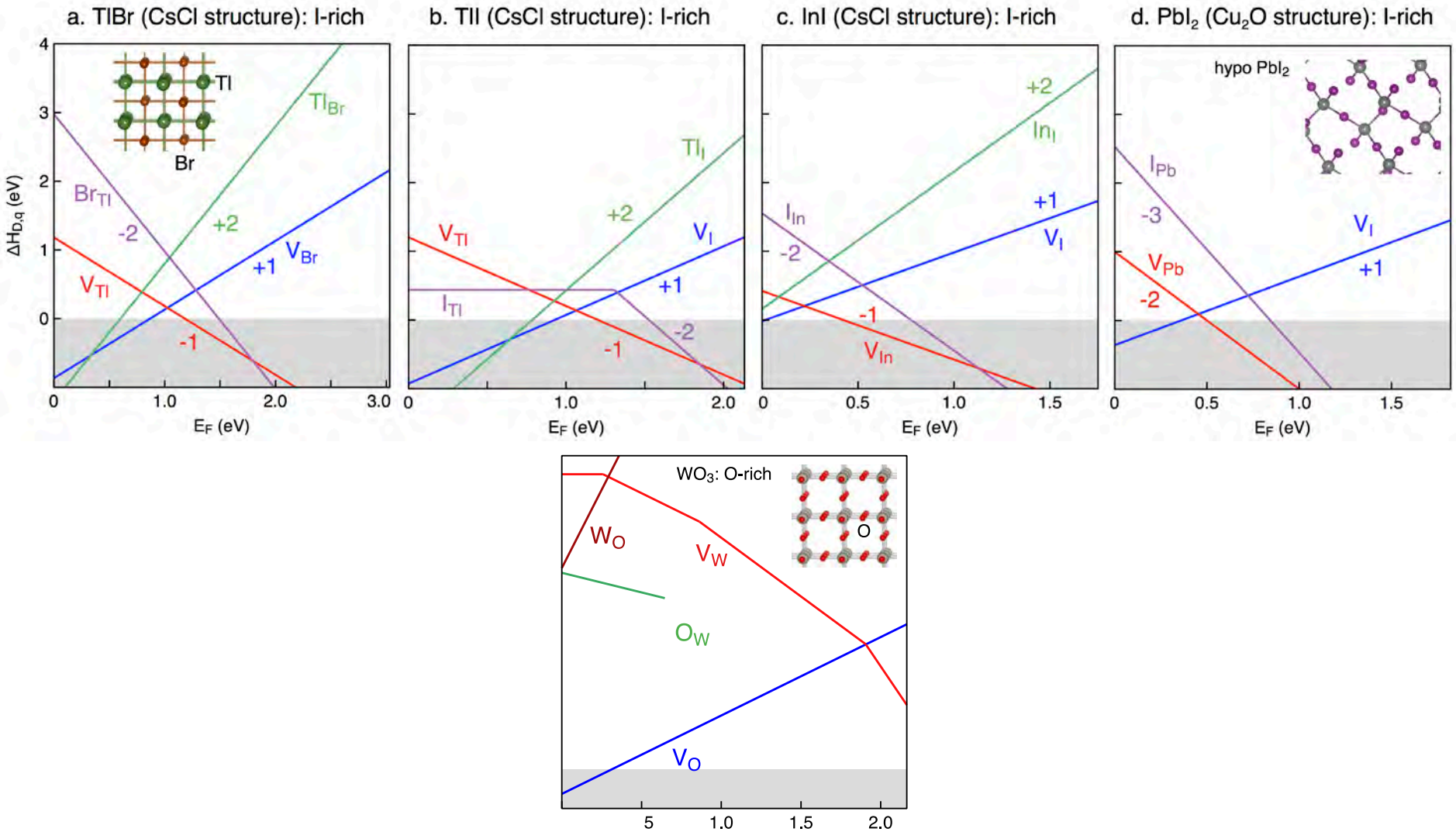
change the **energies** (chemistry)

# Back to those binary iodides



Kurchin, R. C., Gorai, P., Buonassisi, T., & Stevanović, V. (2018). Structural and Chemical Features Giving Rise to Defect Tolerance of Binary Semiconductors. *Chemistry of Materials*, under revision.

# Testing the new theory

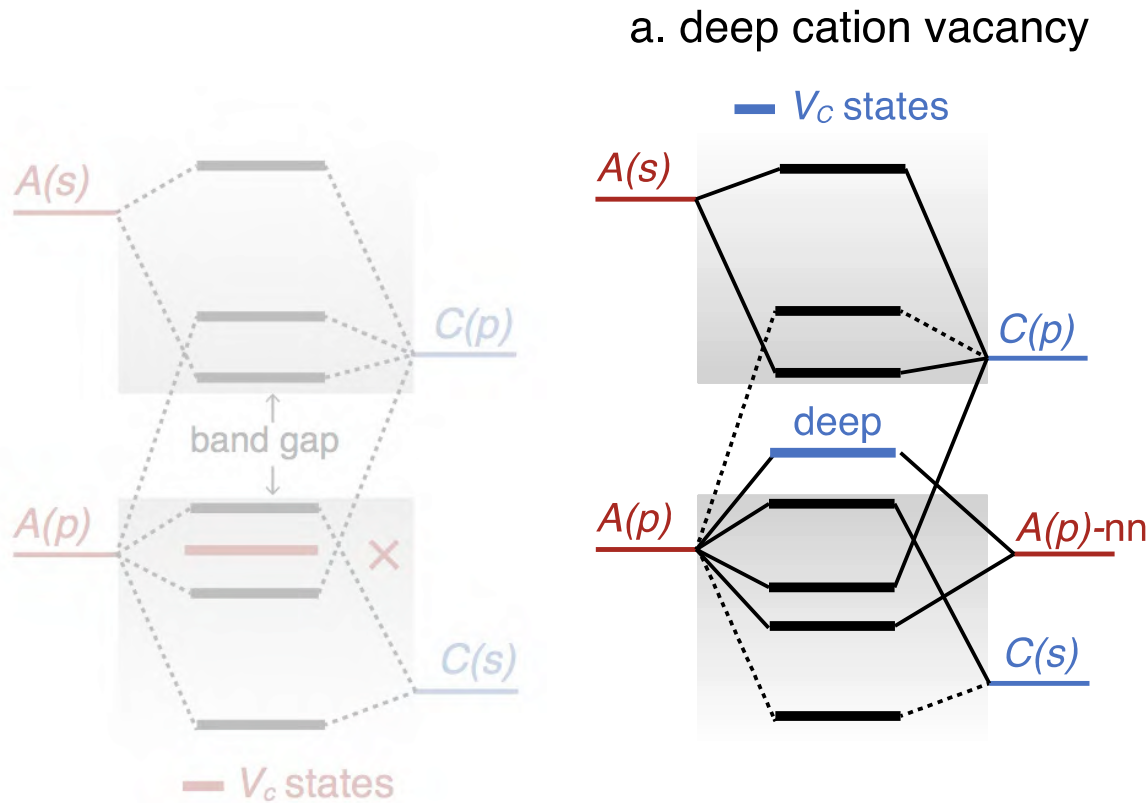


# Thanks, Blue Waters!

- BW Graduate Fellowship gave me intellectual independence and confidence in my research
- And a great community!



# Updated Theory of Defect Tolerance: Cation vacancies (chemistry)



# Updated Theory of Defect Tolerance: Anion vacancies (structure)

