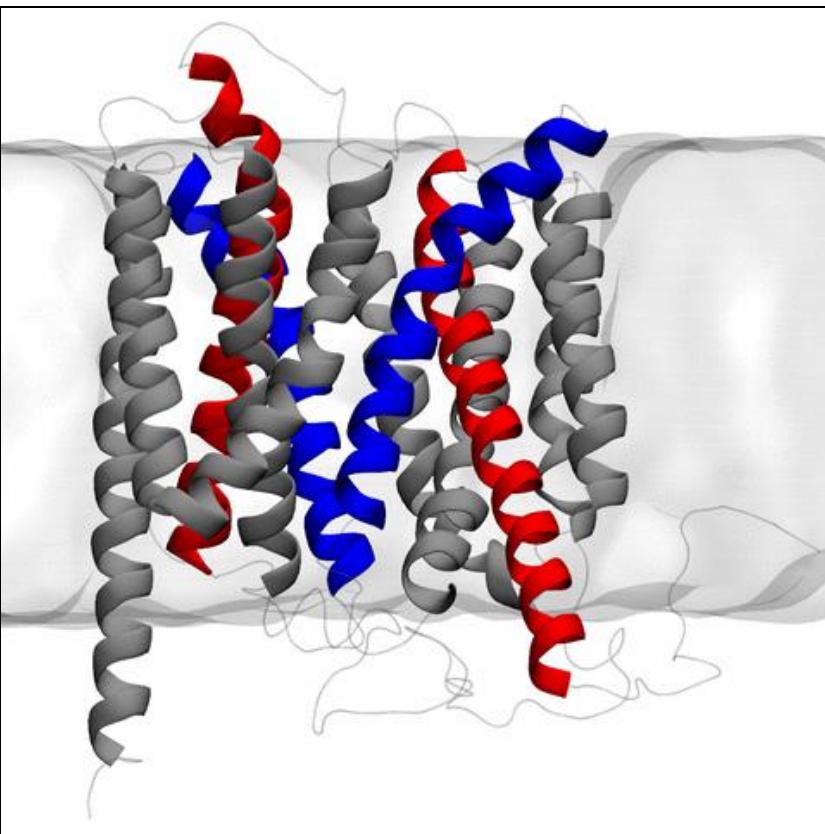


Characterizing Structural Transitions of Membrane Transport Proteins at Atomic Detail



Mahmoud Moradi

PI: Emad Tajkhorshid

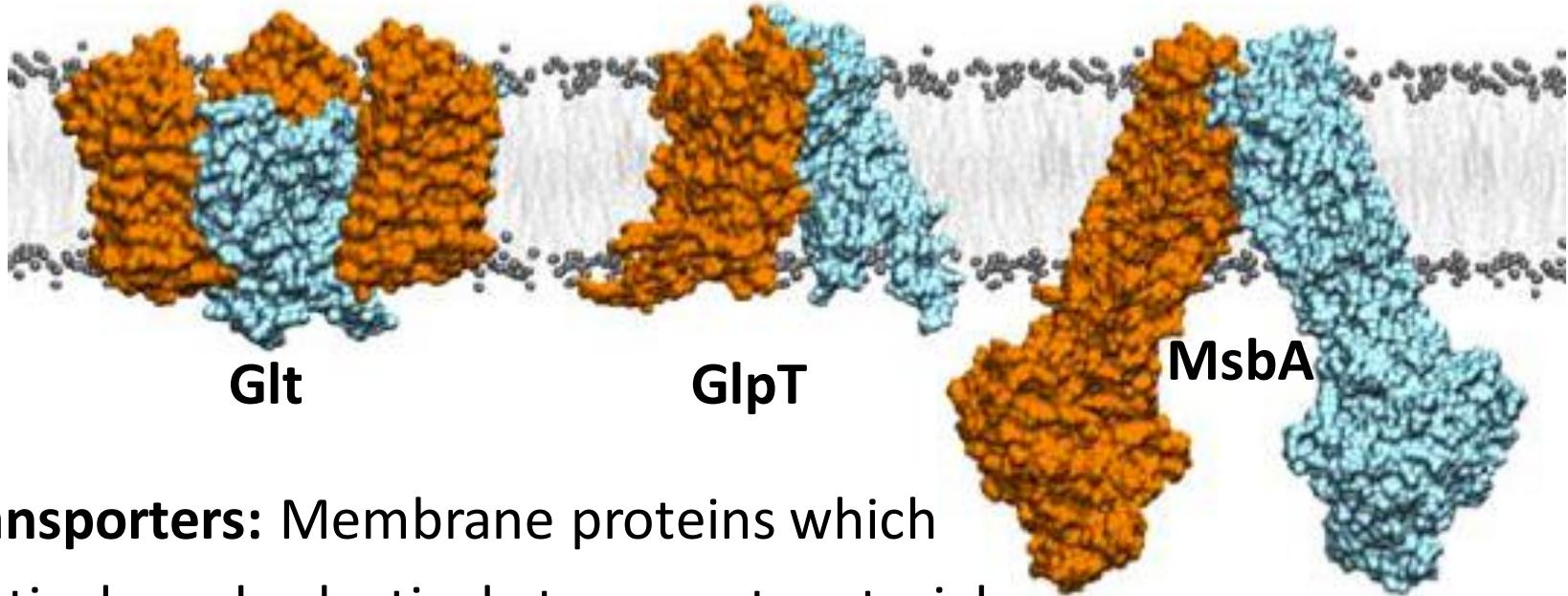
Beckman Institute for
Advance Science and Technology,
University of Illinois at Urbana-Champaign

NCSA Blue Waters Symposium
for Petascale Science and Beyond
Sunriver, Oregon
May 11, 2015

Outline

- **Introduction**
 - GlpT transporter
 - Transport cycle thermodynamics
- **Methodology**
 - Empirical search for reaction coordinates using nonequilibrium simulations
 - Iterative path-finding algorithms and free energy calculations
- **Reconstructed thermodynamic cycle of GlpT**
 - Free energy profile along the cycle
 - Global and local conformational changes and their coupling

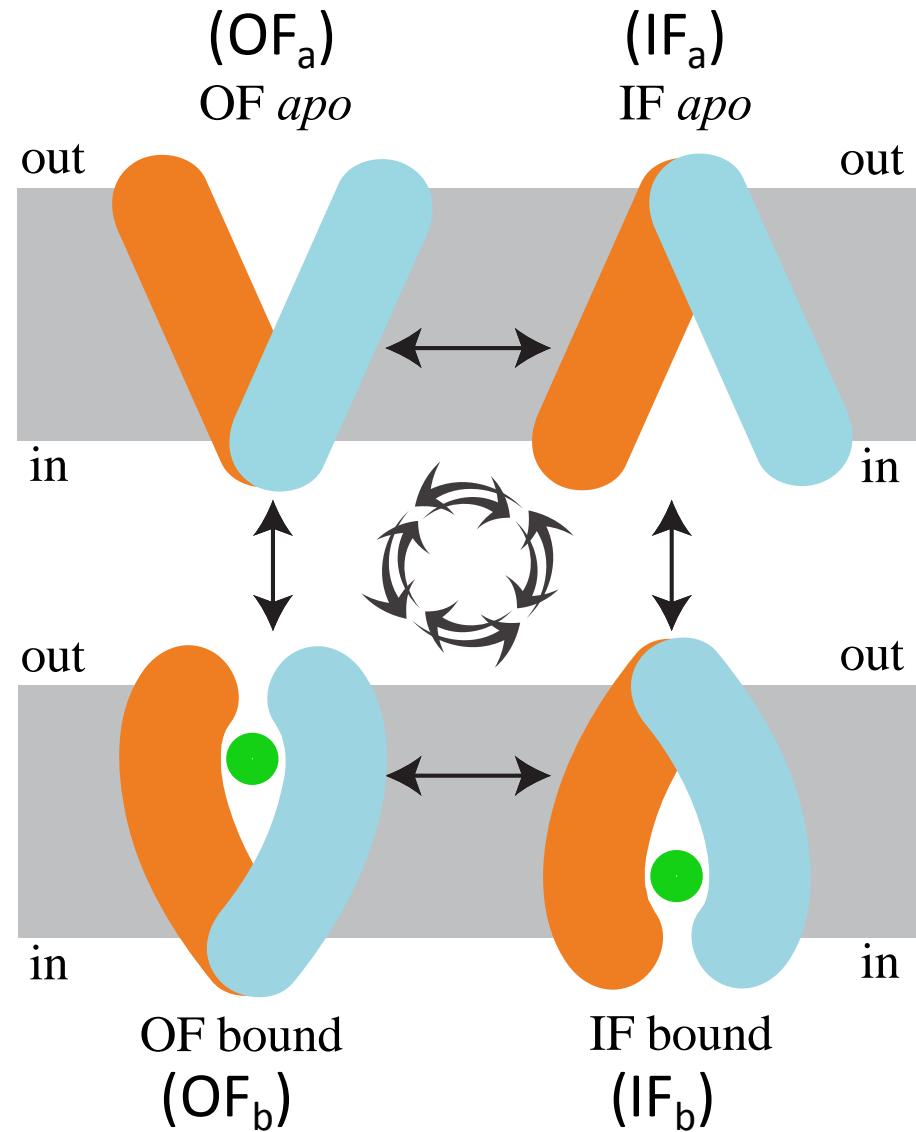
Membrane Transporters



- **Transporters:** Membrane proteins which actively and selectively transport materials (proton, ions, small molecules) across cell membranes.
- **Active transport:** Pumping substrates against their concentration gradient (from low to high concentration).
- **Source of energy:**
 - metabolic energy, e.g. from ATP hydrolysis (**primary**).
 - electrochemical gradient of an ion (**secondary**).

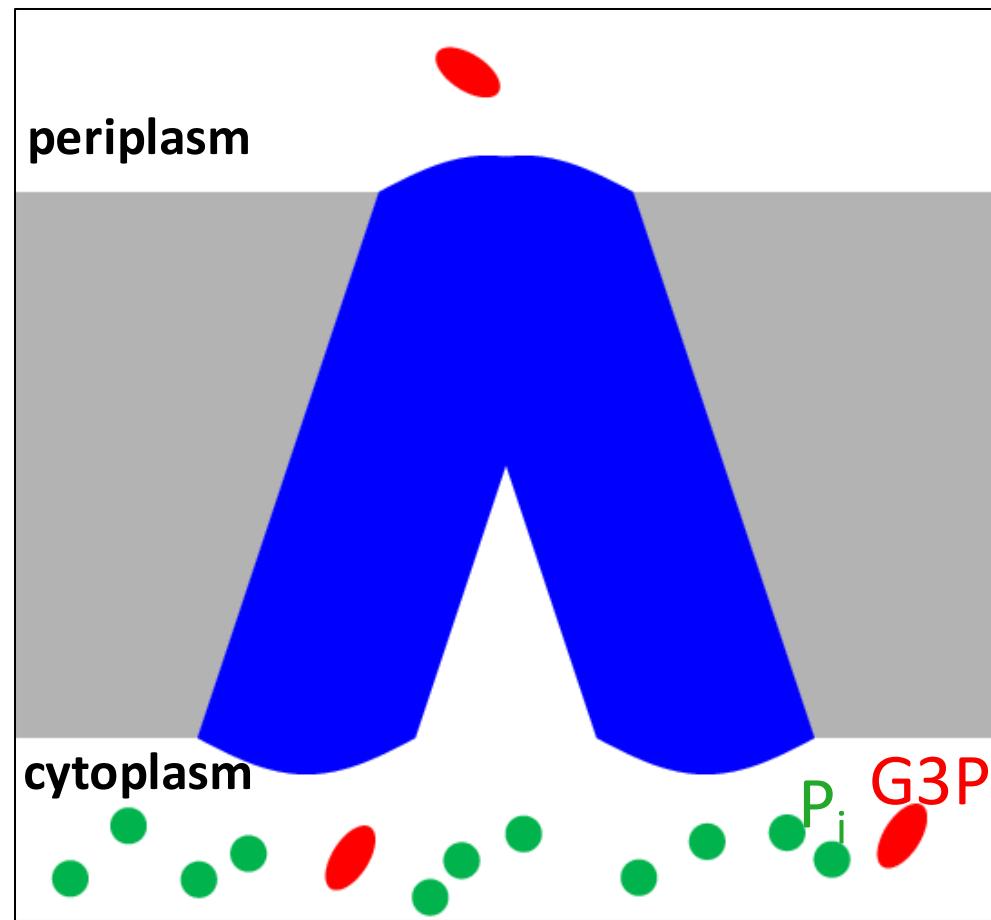
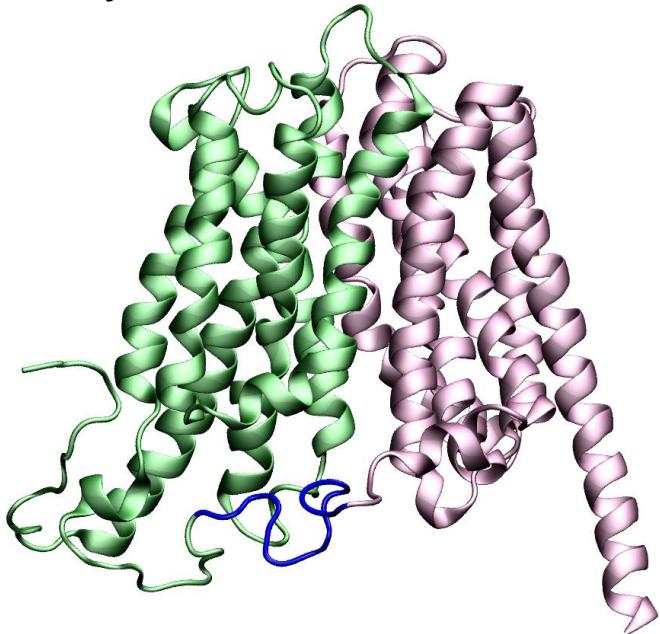
Alternating-Access Mechanism

- Membrane transporters rely on **large-scale conformational changes** to alternate between inward-facing (**IF**) and outward-facing (**OF**) states to pump the substrate against its concentration gradient, without being open (having the binding site accessible) to both sides of the membrane simultaneously.



Glycerol-3-phosphate (G3P) transporter (GlpT)

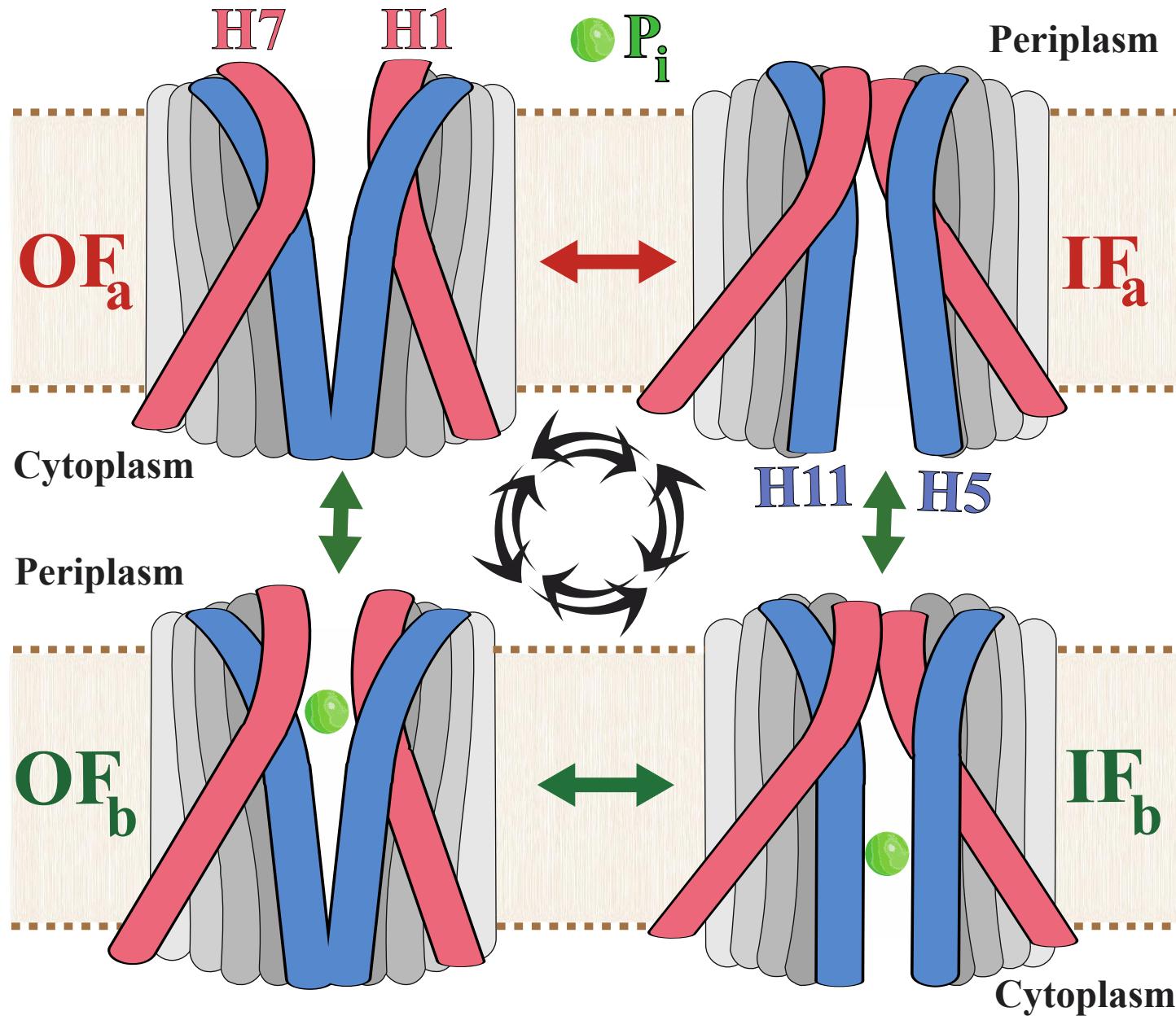
- Major facilitator superfamily (MFS)
- Secondary active transporter
- Crystallized in the IF state.



- GlpT transports **G3P** using **P_i** gradient.
- **P_i:P_i** exchanger (in the absence of organic phosphate)
- **Rate-limiting step: IF-OF** interconversion.

Huang, et al., *Science* 301, 616 (2003).

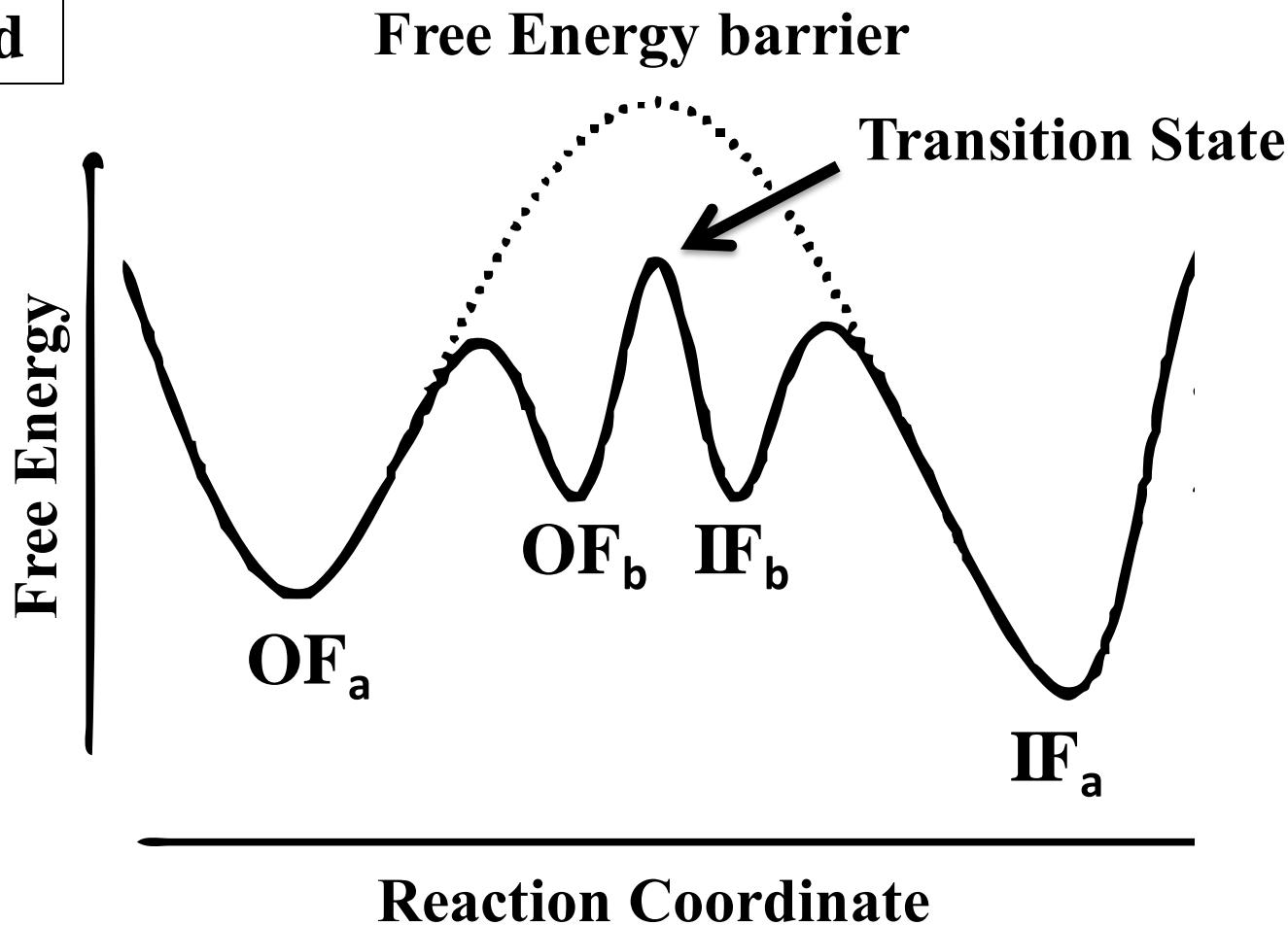
Transport cycle thermodynamics



Transport cycle thermodynamics

a: *apo*

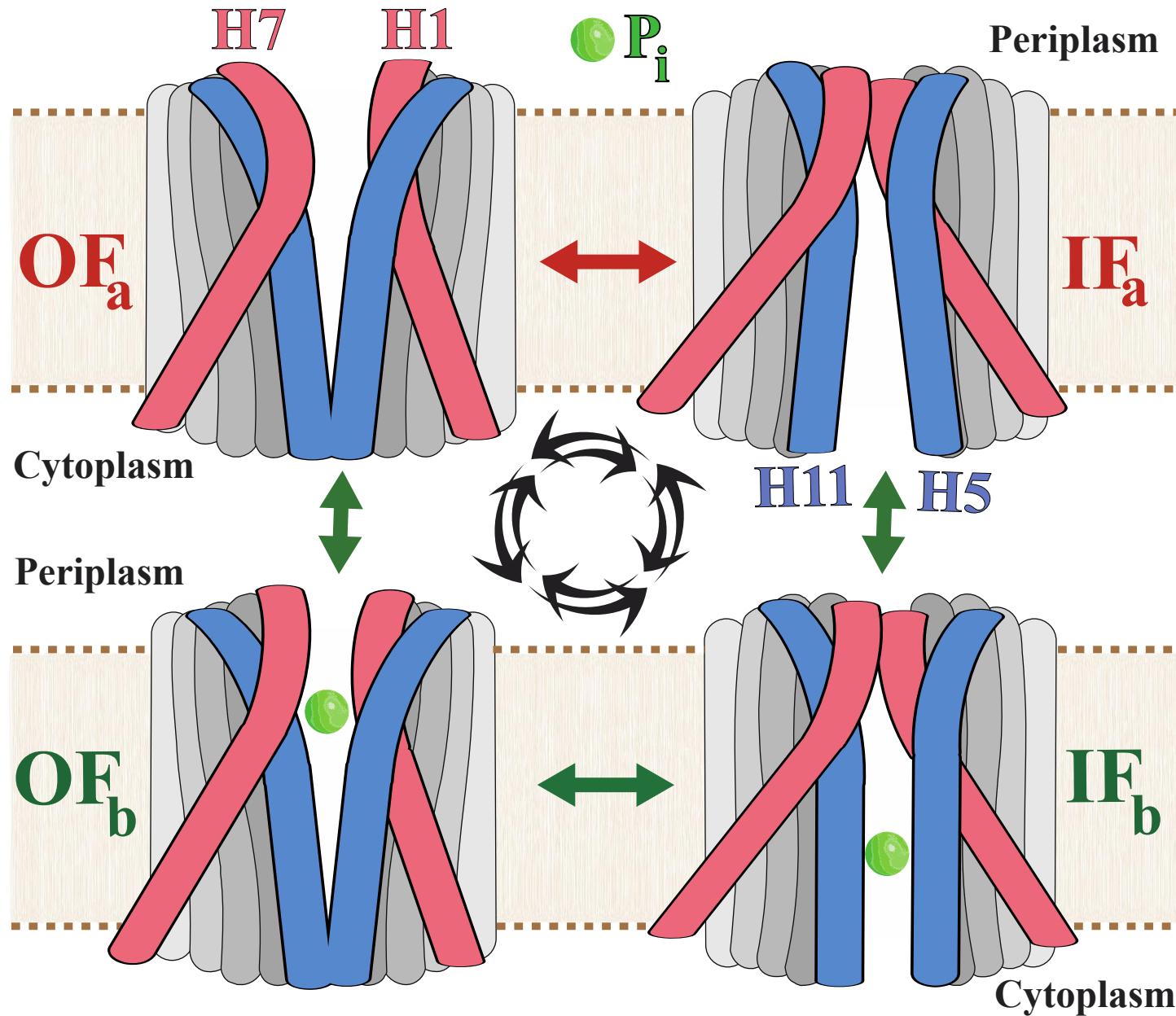
b: bound



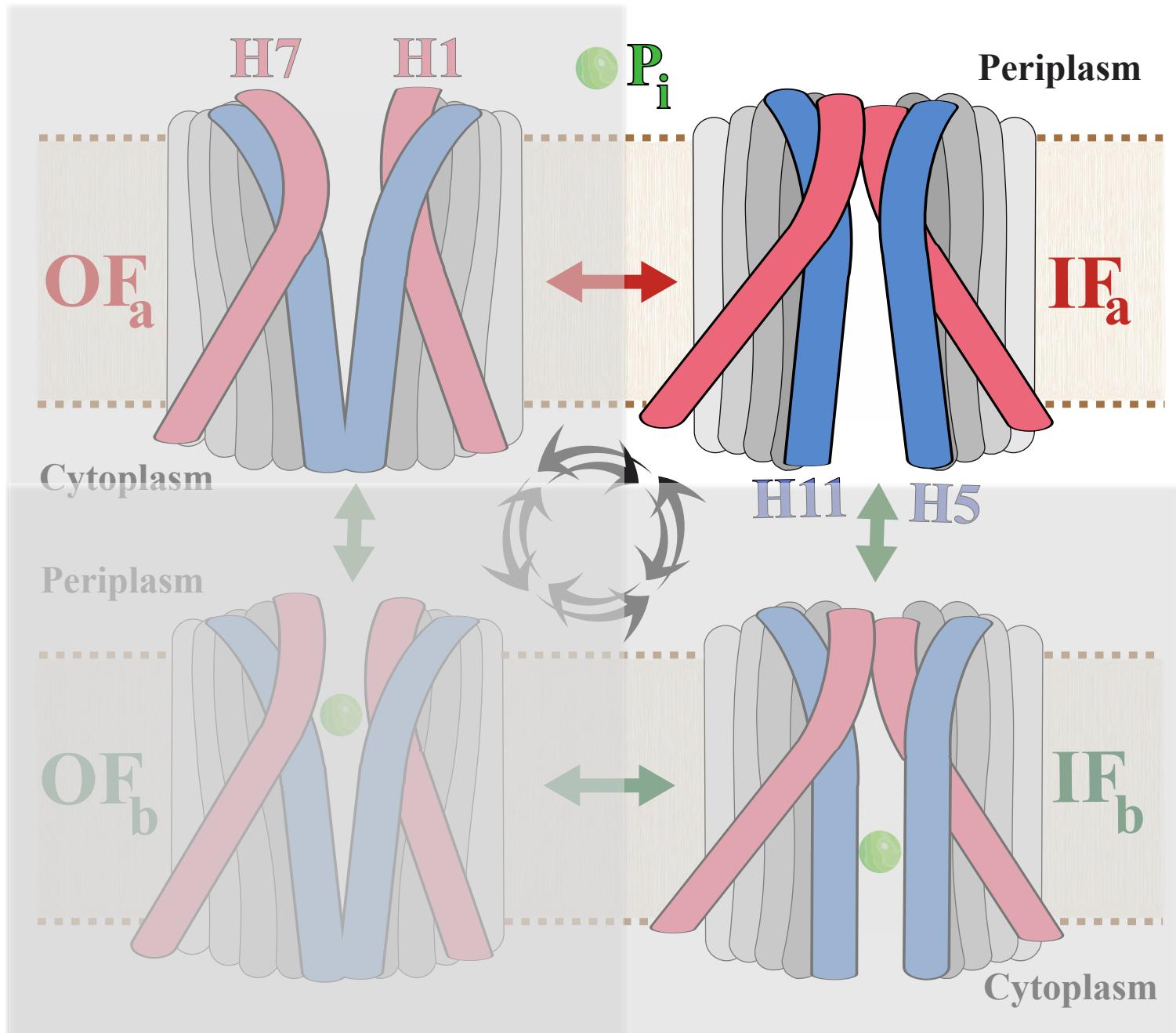
Lemieux, et al., *Curr. Opin. Struct. Biol.* **14**, 405 (2004).

Law, et al., *Biochemistry* **46**, 12190 (2007).

Full thermodynamic cycle



the only available crystal structure



Key Challenge:

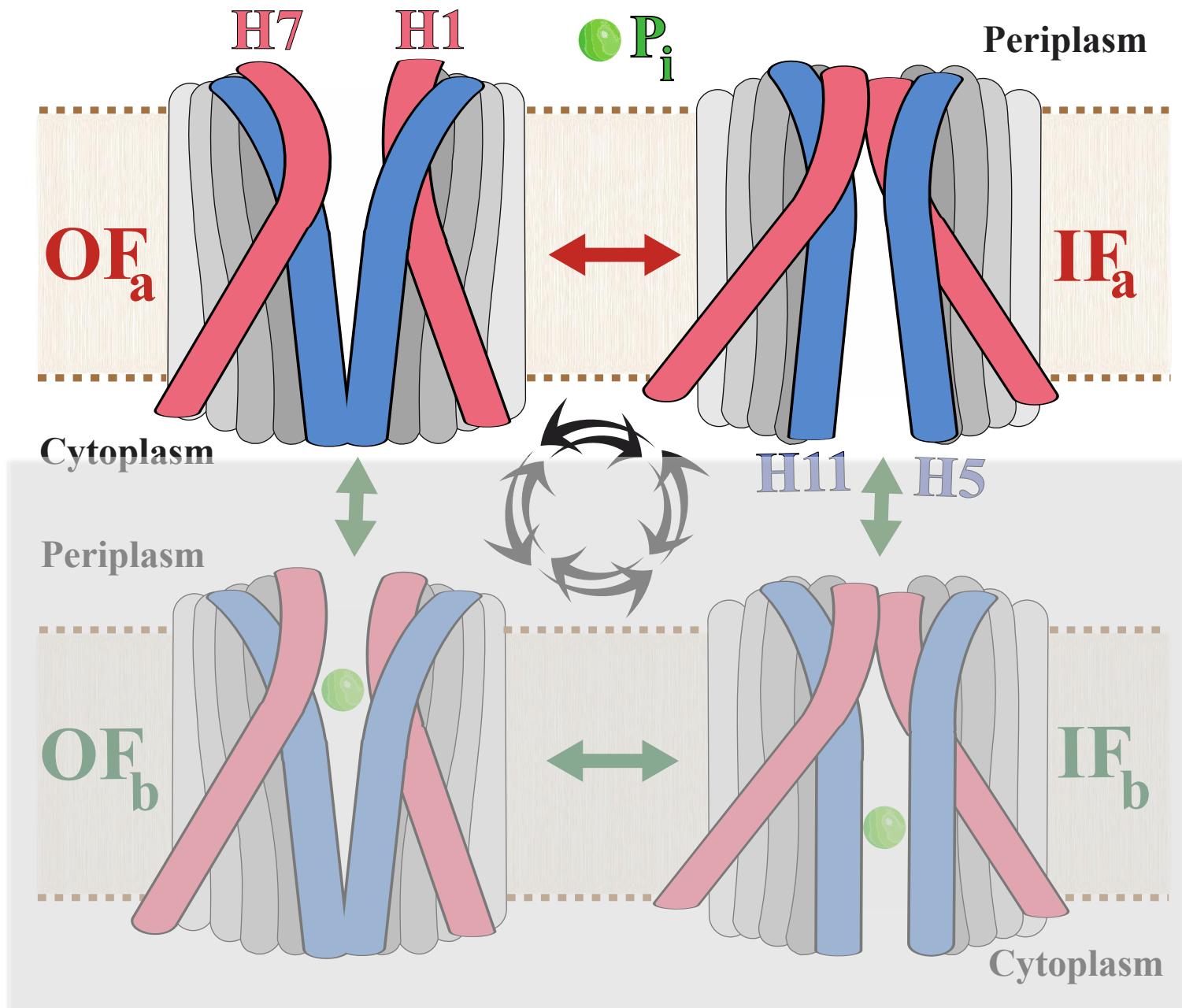
- **Slow dynamics**
 - Timescale gap between feasible all-atom molecular dynamics (MD) simulations and actual functionally relevant biomolecular processes.

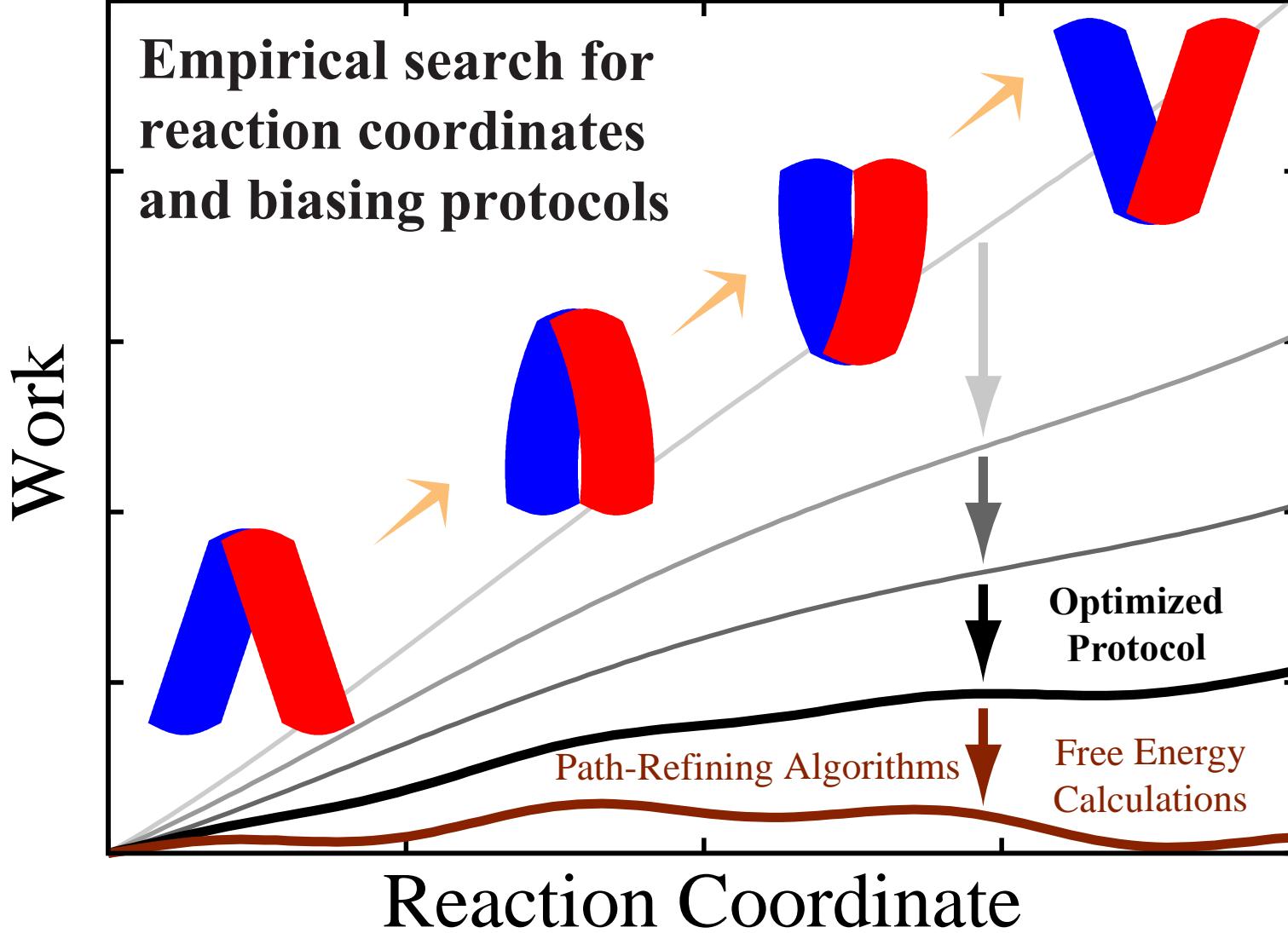
Sampling Strategies:

- **Long simulation**
 - application-specific computers
- **Multiple-copy simulations**
 - distributed computing
- **Enhanced sampling**
 - biased/adaptive simulations

**Loosely-coupled
multiple-copy algorithms
(petascale computing)**

Step 1: $\text{OF}_a \leftrightarrow \text{IF}_a$





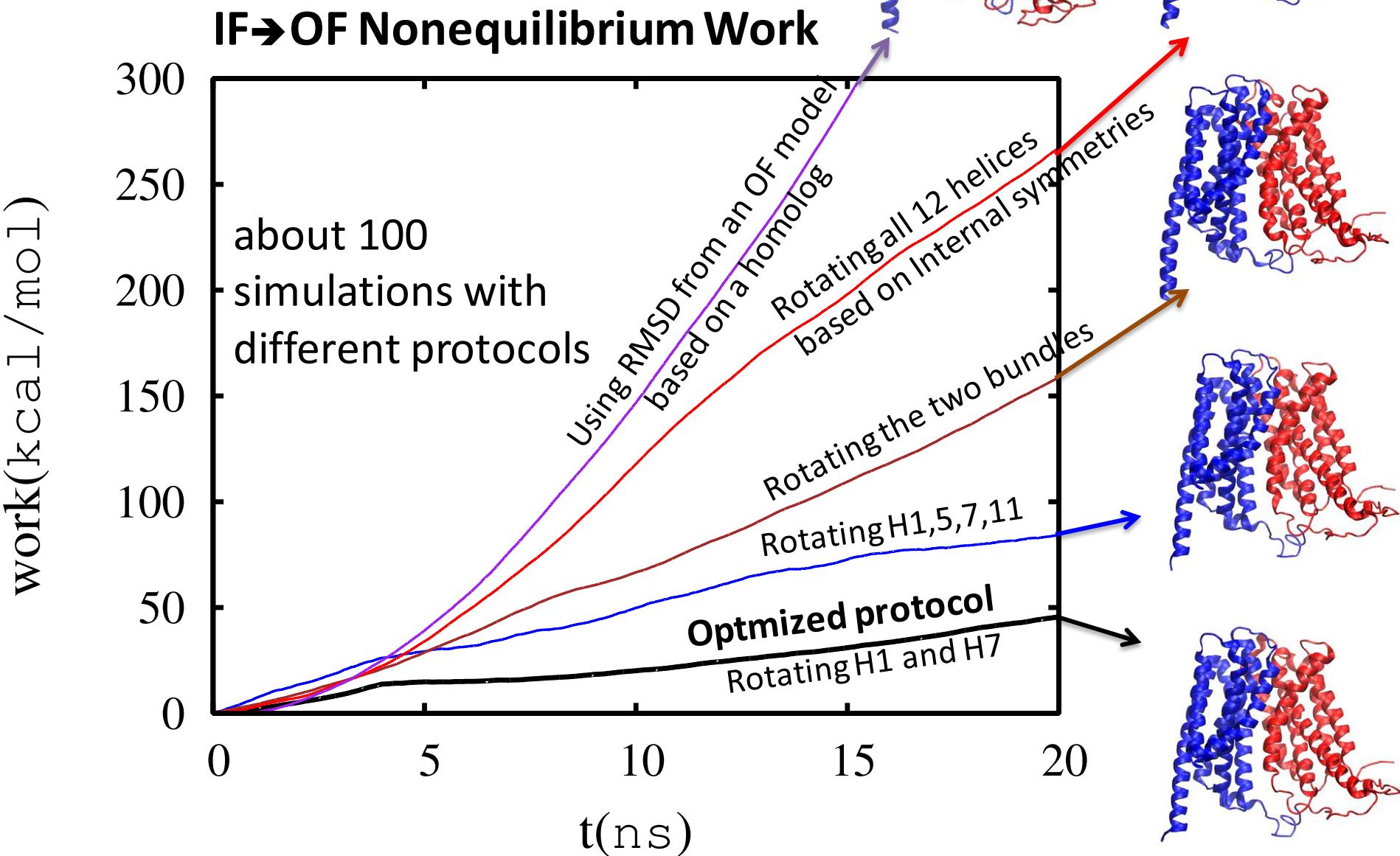
Theory/Method:

- Moradi et al., CPL 518 109 (2011)
Moradi et al., JCP 140 034114,5 (2014)
Moradi et al., JCTC 10 2866 (2014)

Application:

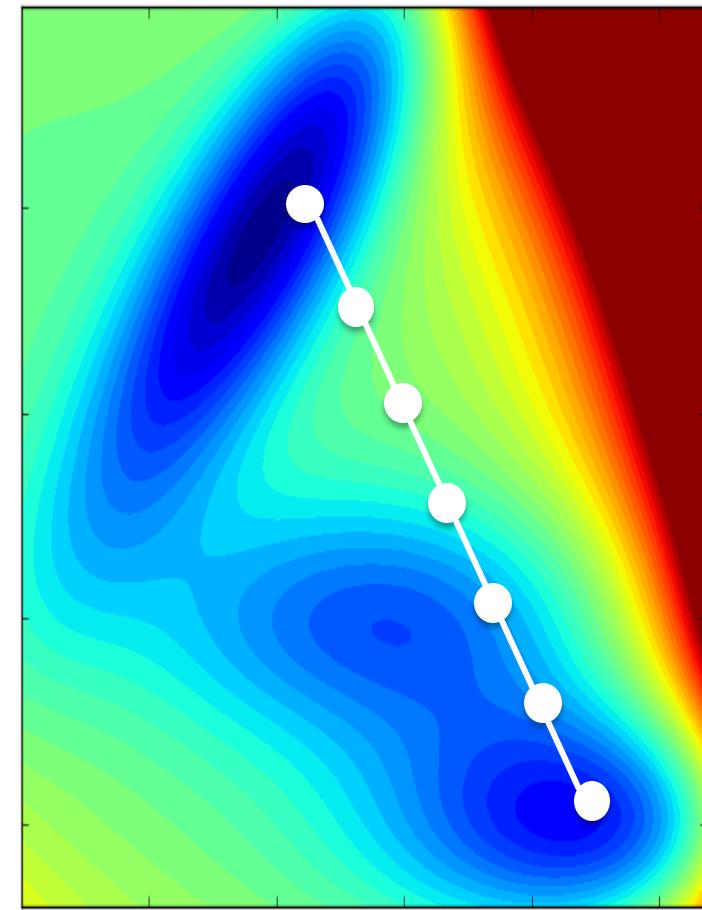
- Moradi et al., PNAS 106 20746 (2009)
Moradi et al., NAR 41 33 (2013)
Moradi et al., PNAS 110 18916 (2013)

Empirical search for reaction coordinates and biasing protocols:



Path-finding algorithms

- String Method (finding approximate minimum free energy pathways on high-dimensional spaces)
 - A pathway is represented by a “string”, i.e., an ordered series of images $\{\xi_i\}$ connecting reactant and product regions.
 - The string is iteratively updated according to some “rule” until converges to a stationary solution:
$$\dot{\xi}(s) \parallel g^{-1}(\xi) \nabla_{\xi} F(\xi)$$
- Maragliano, Fischer, Vanden-Eijnden, and Ciccotti
J. Chem. Phys. 2006, 125, 024106.
- Ren, Vanden-Eijnden, Maragakis, and E
J. Chem. Phys. 2005, 123, 134109.
- Vanden-Eijnden and Venturoli; J. Chem. Phys. 2009, 130, 194103.



Path-finding algorithms

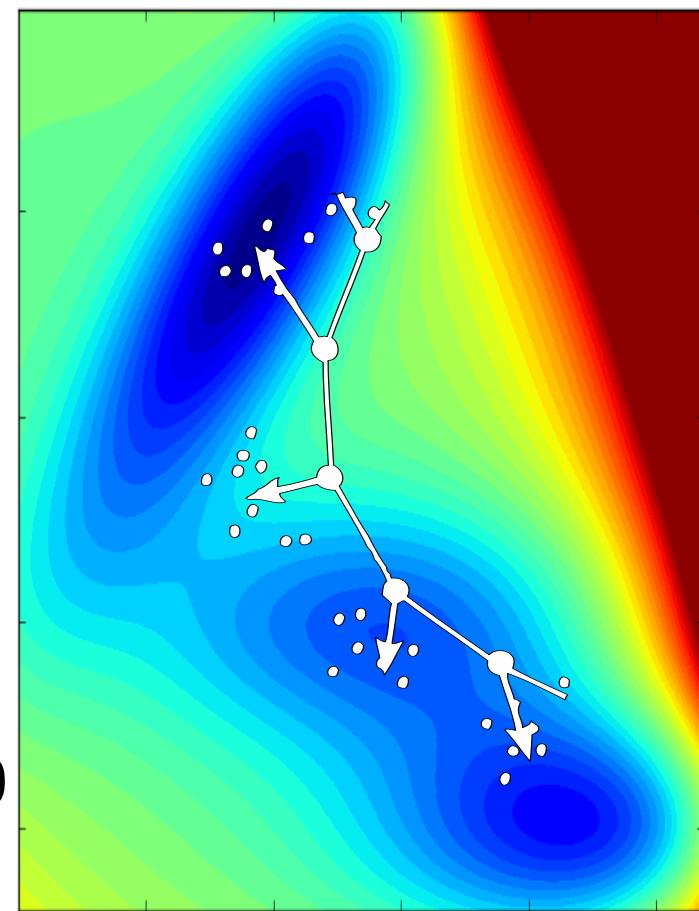
- String Method with Swarms of Trajectories (SMwST):

- For each image **tens of copies** are launched:
- Start with an initial string $\{\xi_i\}$
- (1) Restrain M copies of each image at the current ξ_i
- (2) Release the restraint
- (3) Update the centers: $\xi_i = \langle \xi_i^t \rangle$
- (4) Reparametrize

Pan, Sezer, and Roux
J. Phys. Chem. B 2008,
112, 3432–3440.

Collective variables: $\{Q\} =$
 $\{Q_1, Q_2, \dots, Q_{12}\}$

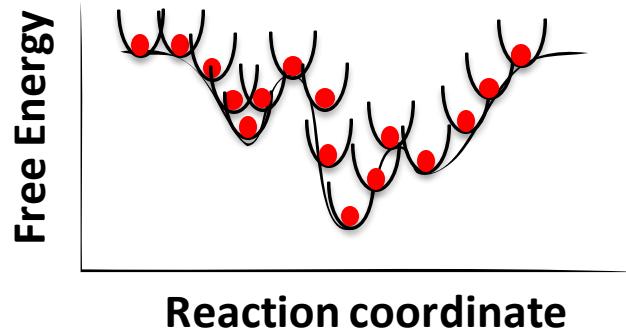
Number of replicas: $50 \times 20 = 1000$
Simulation time: 1 ns/replica



Free energy calculations

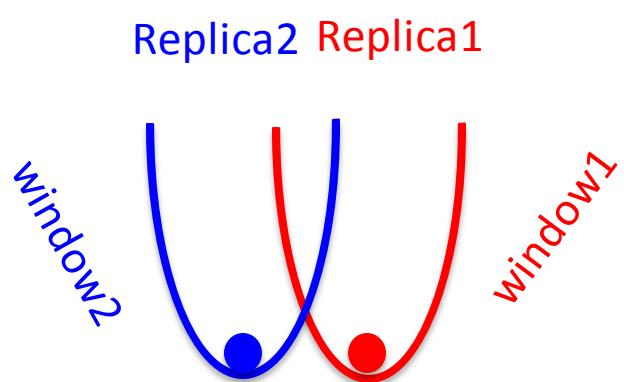
- Bias-exchange umbrella sampling (BEUS)
(Loosely coupled multiple-copy MD)
 - Umbrella sampling

$$U_B(\chi_i^t) = \frac{1}{2} k (\chi_i^t - \bar{\chi}_i)^2$$



- Replica-exchange MD

$$p(x_1x_2 \rightarrow x_2x_1) = \min \left(1, \frac{\pi_1(x_2)\pi_2(x_1)}{\pi_1(x_1)\pi_2(x_2)} \right)$$
$$\min \left(1, \frac{e^{-\beta U_1(\xi_2)} e^{-\beta U_2(\xi_1)}}{e^{-\beta U_1(\xi_1)} e^{-\beta U_2(\xi_2)}} \right)$$



Iterative path-refining algorithms and free energy calculations

BEUS

**Bias-Exchange Umbrella Sampling
(Free Energy Calculation)**

MCA

PHSM

**Post-Hoc String Method
(Path-Finding Algorithm)**

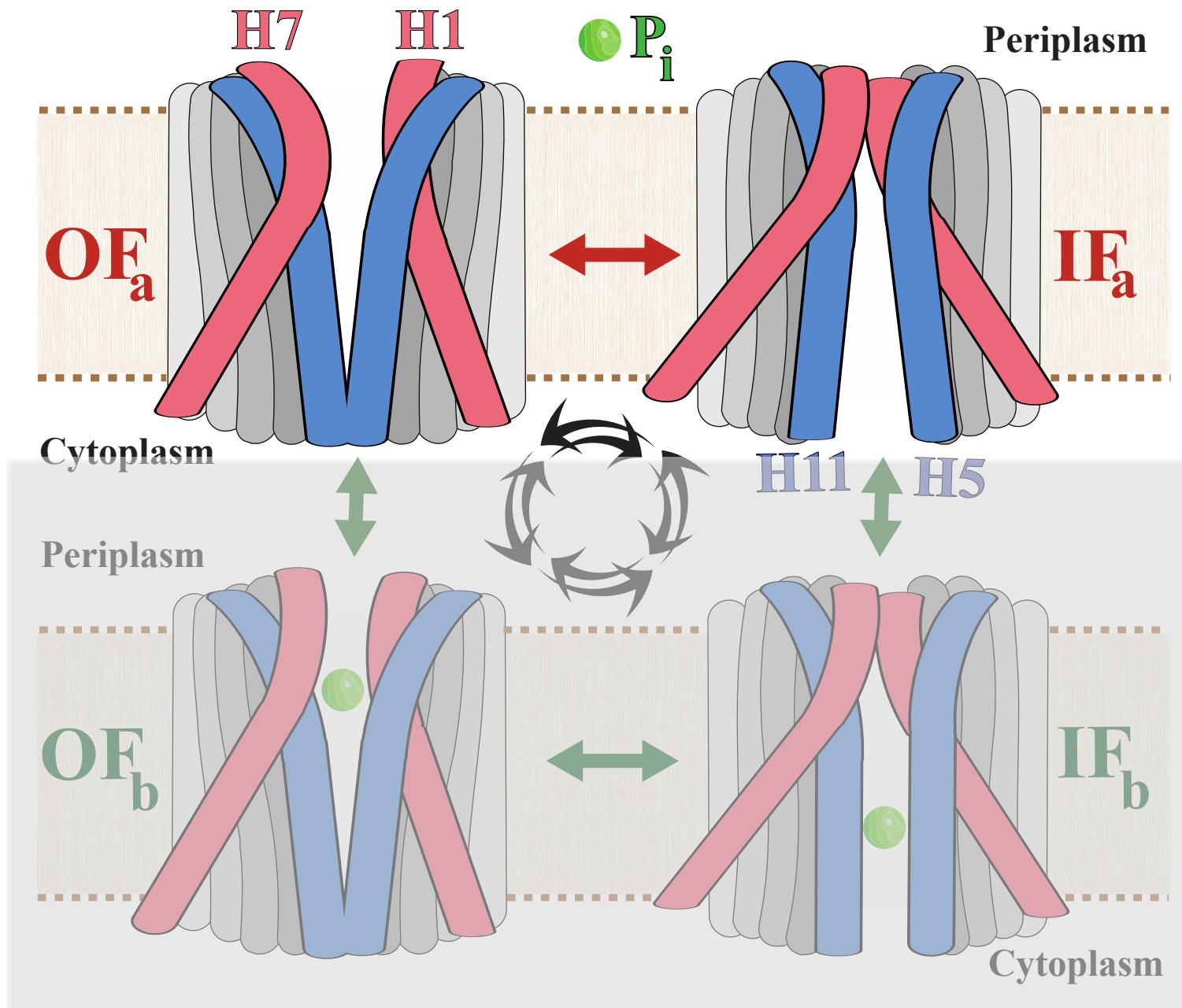
Analysis
Technique

SMwST

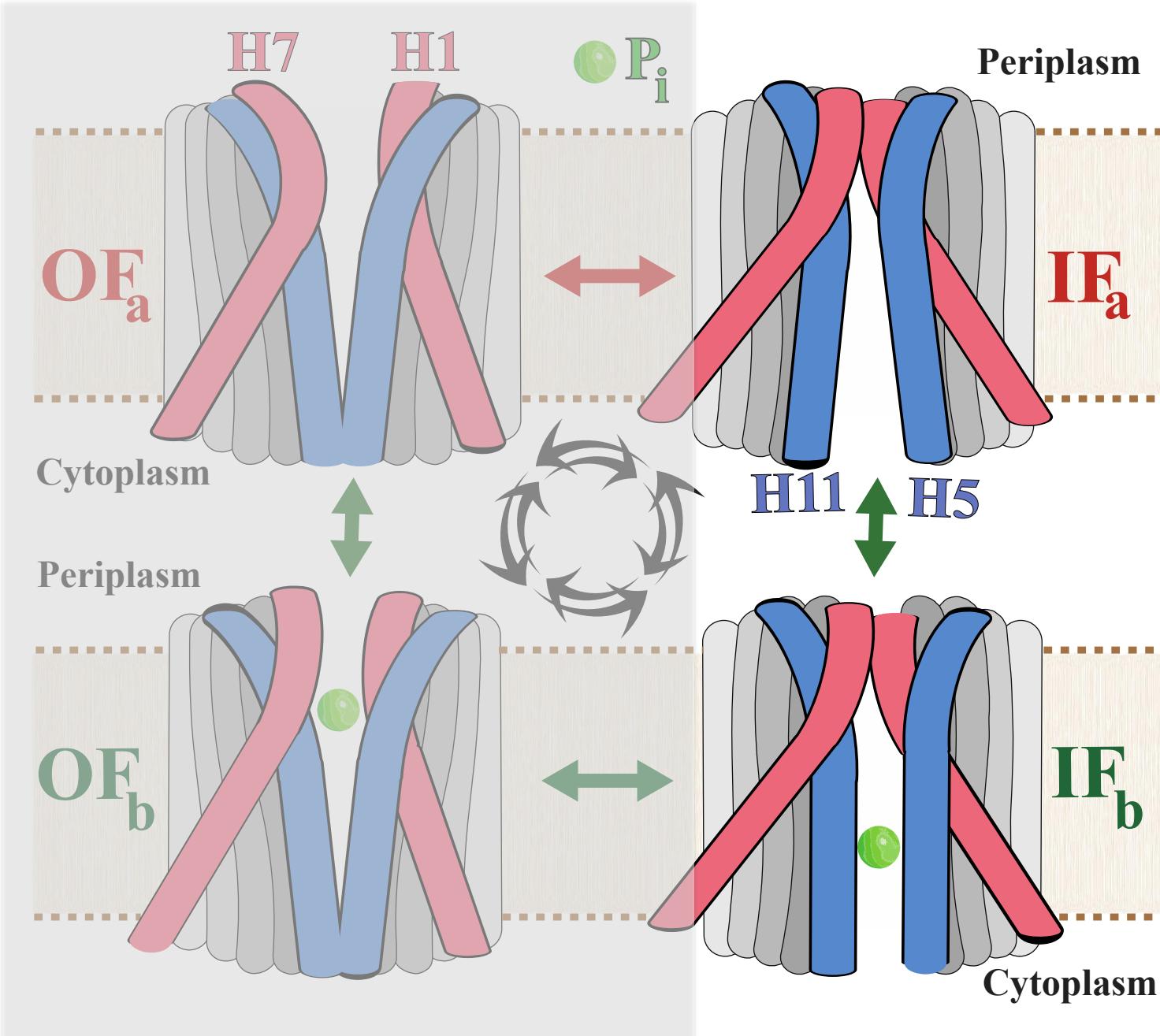
MCA

**String Method with Swarms of Trajectories
(Path-Finding Algorithm)**

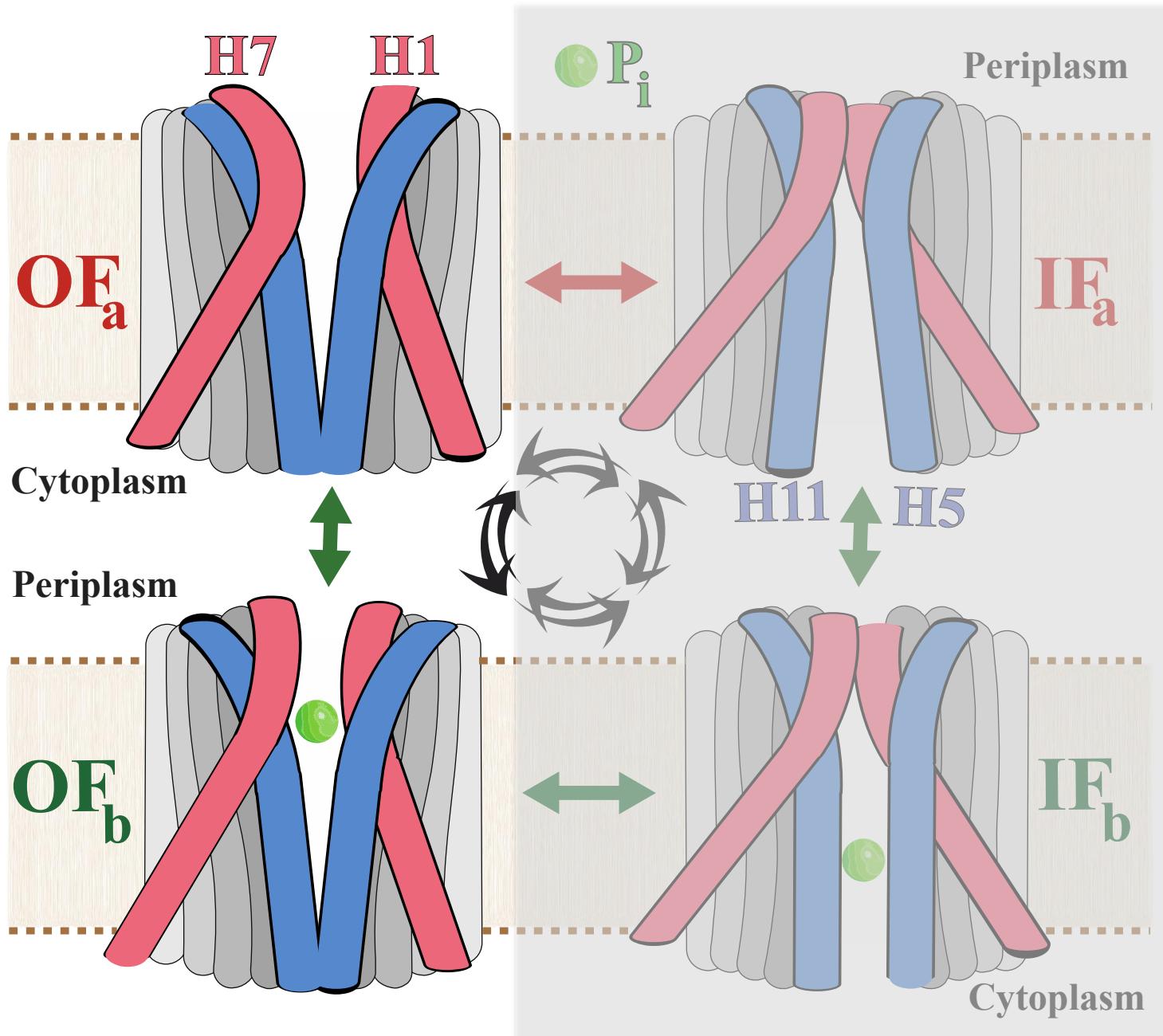
Step 1: $\text{OF}_a \leftrightarrow \text{IF}_a$



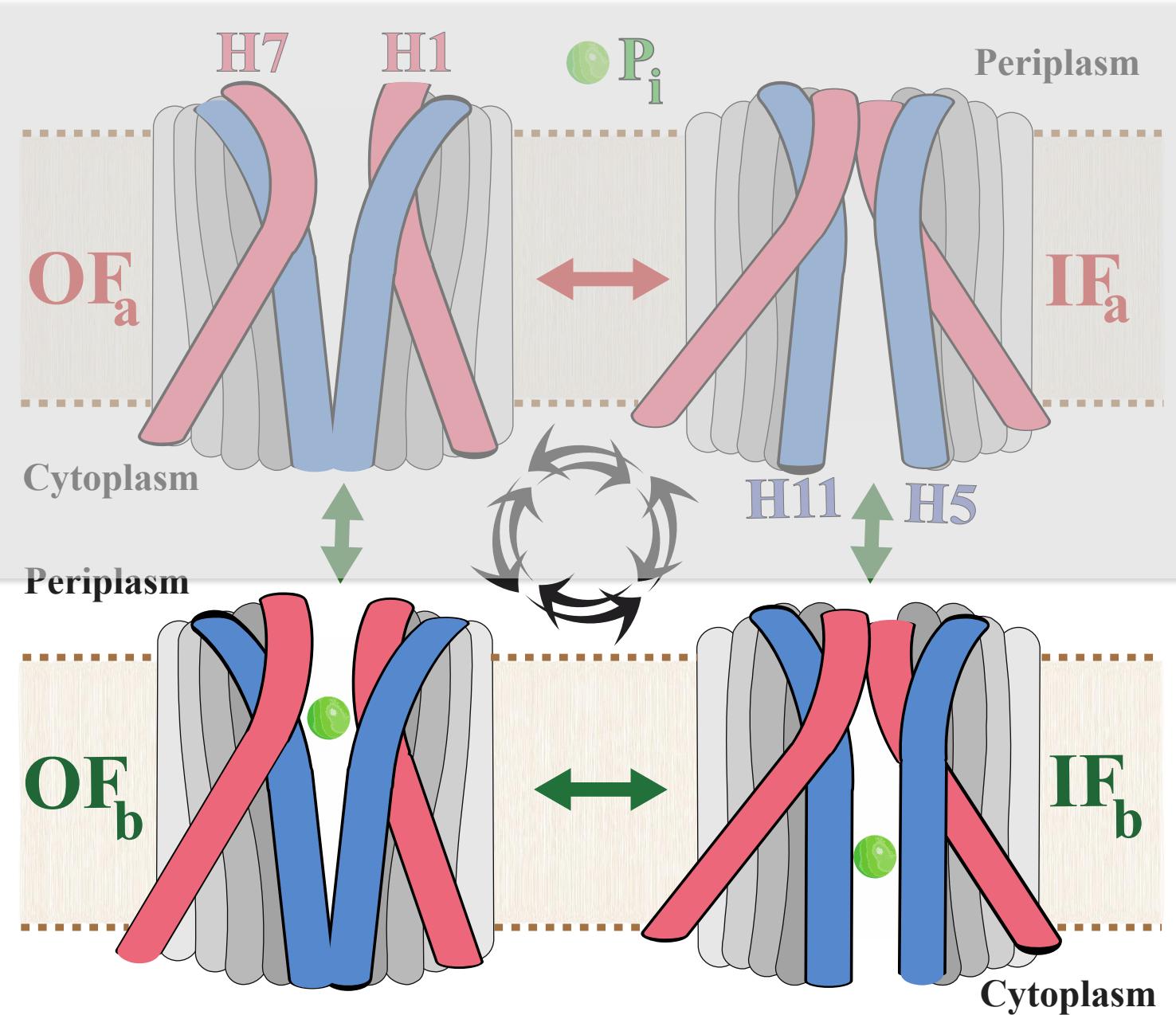
Step 2: $\text{IF}_a \leftrightarrow \text{IF}_b$



Step 3: $\text{OF}_a \leftrightarrow \text{OF}_b$

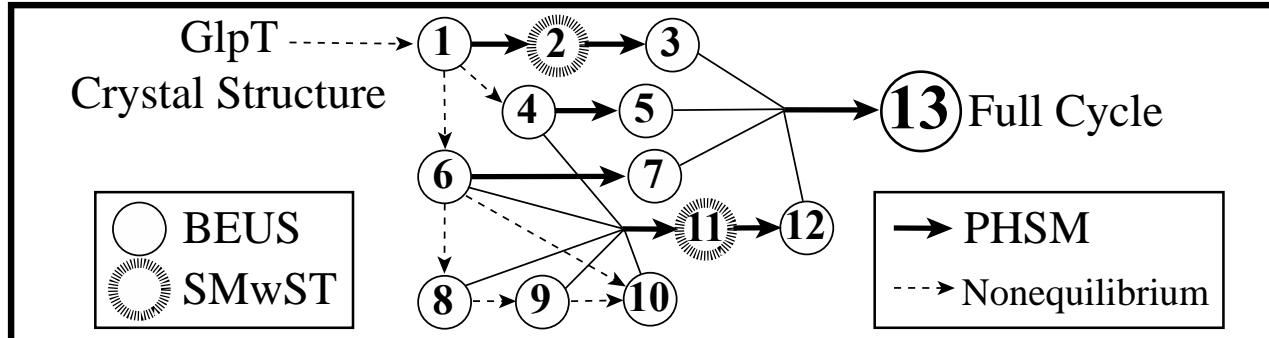


Step 4: $\text{OF}_b \leftrightarrow \text{IF}_b$

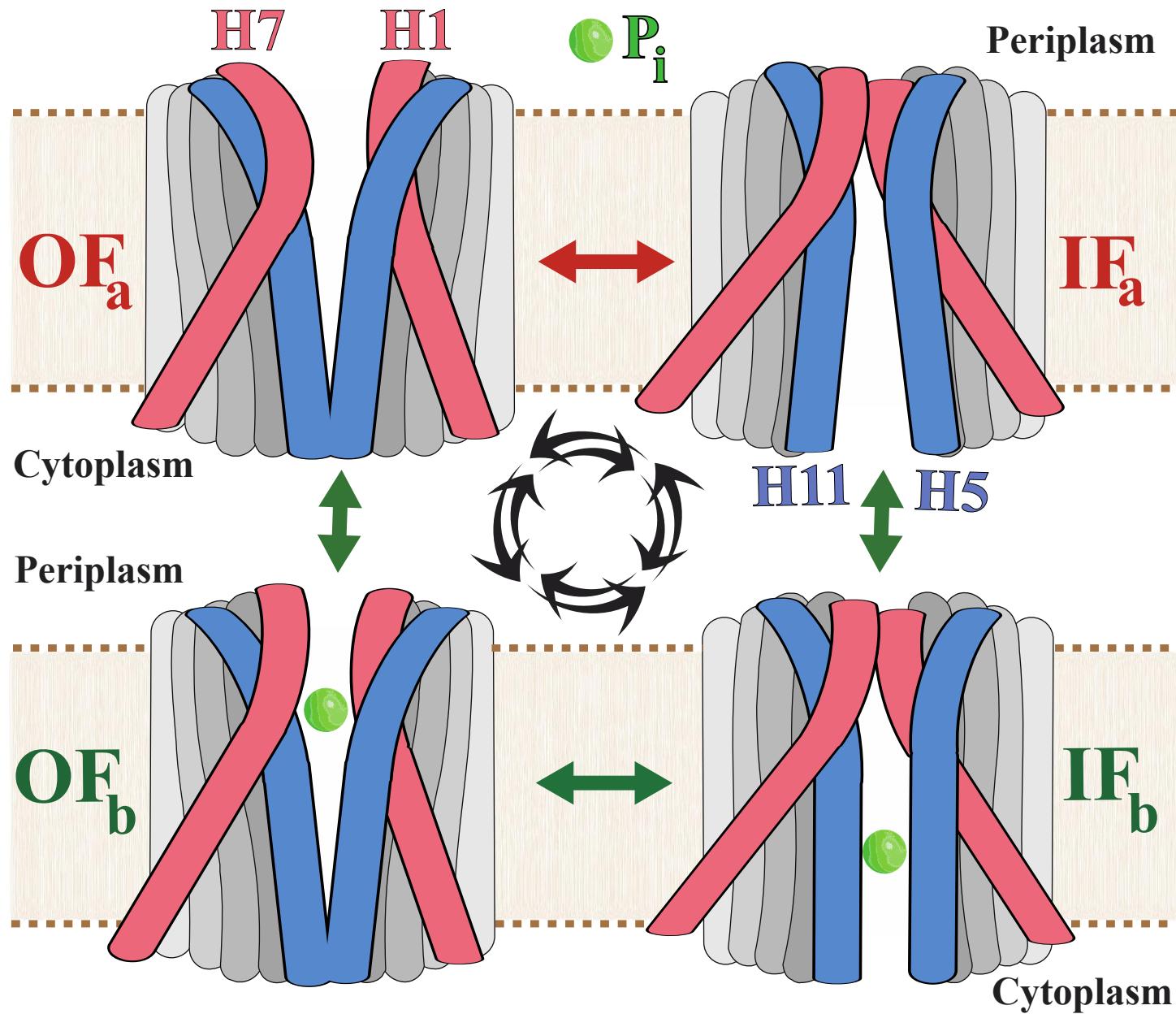


Simulation protocols

Transition	Technique	Collective Variables	# of Replicas × Runtime	=	Runtime
1	$\text{IF}_a \leftrightarrow \text{OF}_a$	BEUS	(Q_1, Q_7)	12×40 ns	= 0.5 ms
2		SMwST	{ Q }	1000×1 ns	= 1 ms
3		BEUS	{ Q }	50×20 ns	= 1 ms
4	$\text{IF}_a \leftrightarrow \text{IF}_b$	BEUS	Z_{P_i}	30×40 ns	= 1.2 ms
5		BEUS	($\{Q\}, Z_{P_i}$)	30×40 ns	= 1.2 ms
6	$\text{OF}_a \leftrightarrow \text{OF}_b$	BEUS	Z_{P_i}	30×40 ns	= 1.2 ms
7		BEUS	($\{Q\}, Z_{P_i}$)	30×40 ns	= 1.2 ms
8	$\text{IF}_b \leftrightarrow \text{OF}_b$	BEUS	(Q_1, Q_7)	24×20 ns	= 0.5 ms
9		BEUS	Z_{P_i}	15×30 ns	= 0.5 ms
10		2D BEUS	($\Delta\text{RMSD}, Z_{P_i}$)	200×5 ns	= 1 ms
11		SMwST	($\{Q\}, Z_{P_i}$)	1000×1 ns	= 1 ms
12		BEUS	($\{Q\}, Z_{P_i}$)	50×20 ns	= 1 ms
13	Full Cycle	BEUS	($\{Q\}, Z_{P_i}$)	150×50 ns	= 7.5 μs
Total Simulation Time					18.7 μs



Each replica consists of ~150,000 atoms



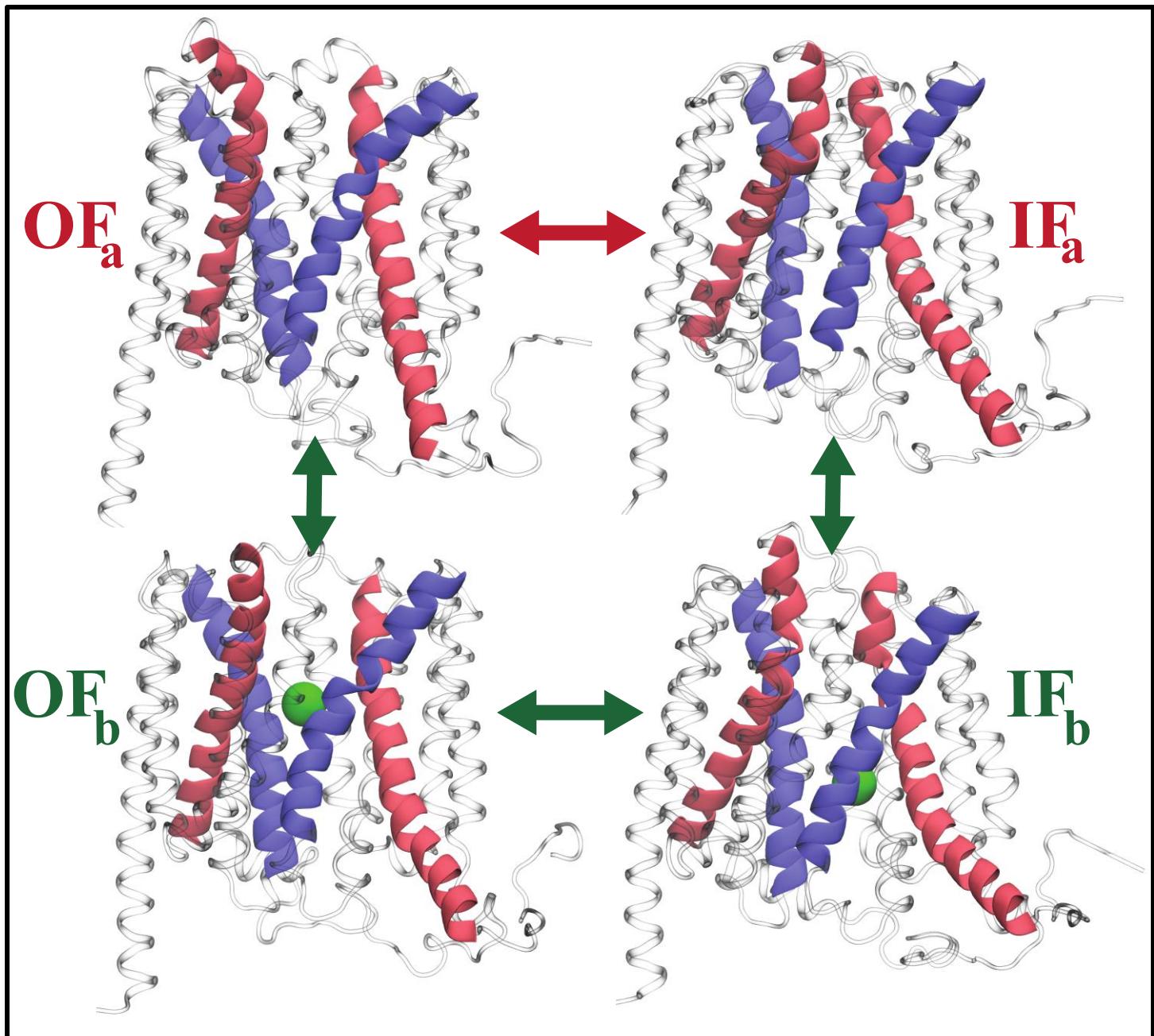


Image Index

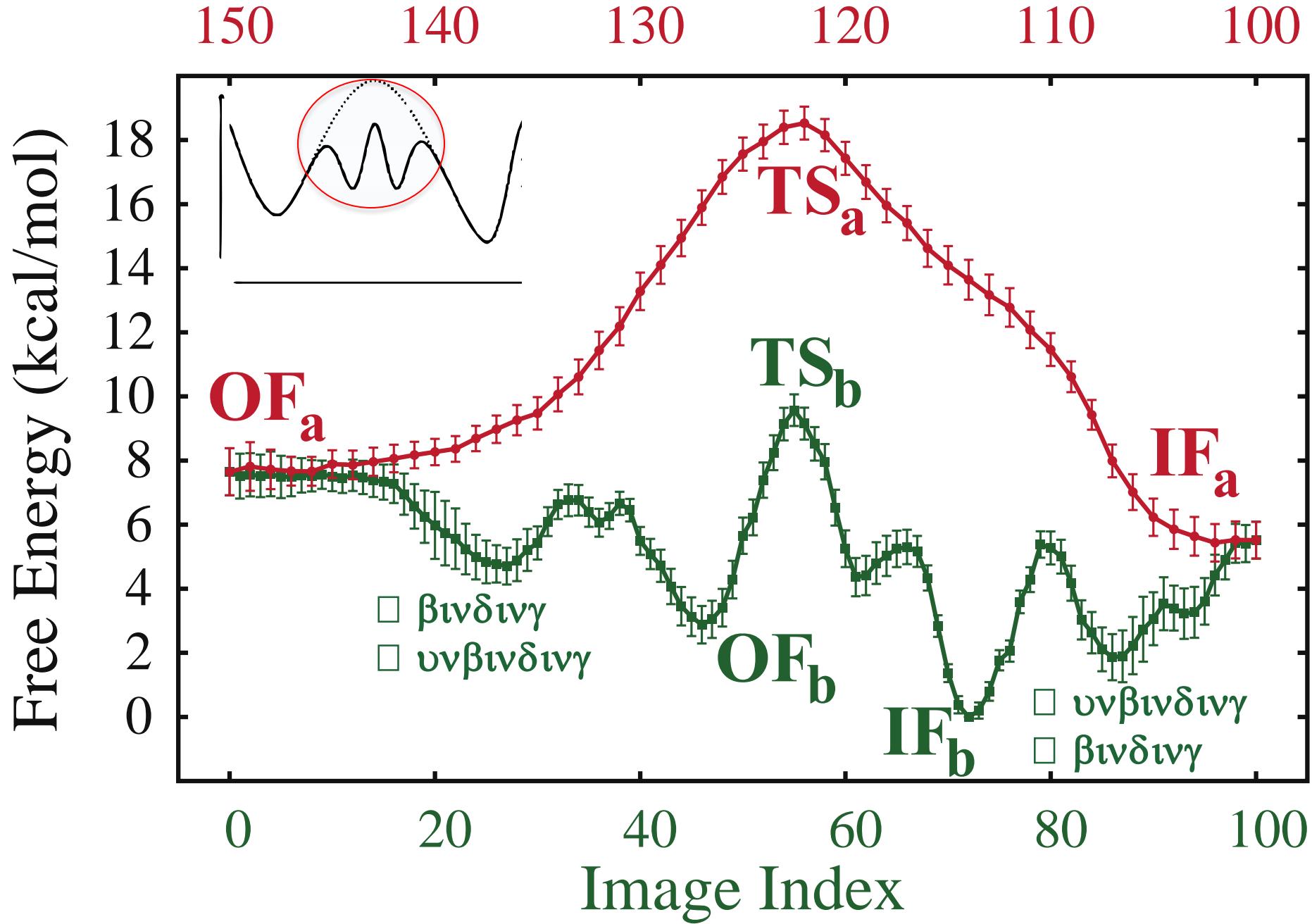
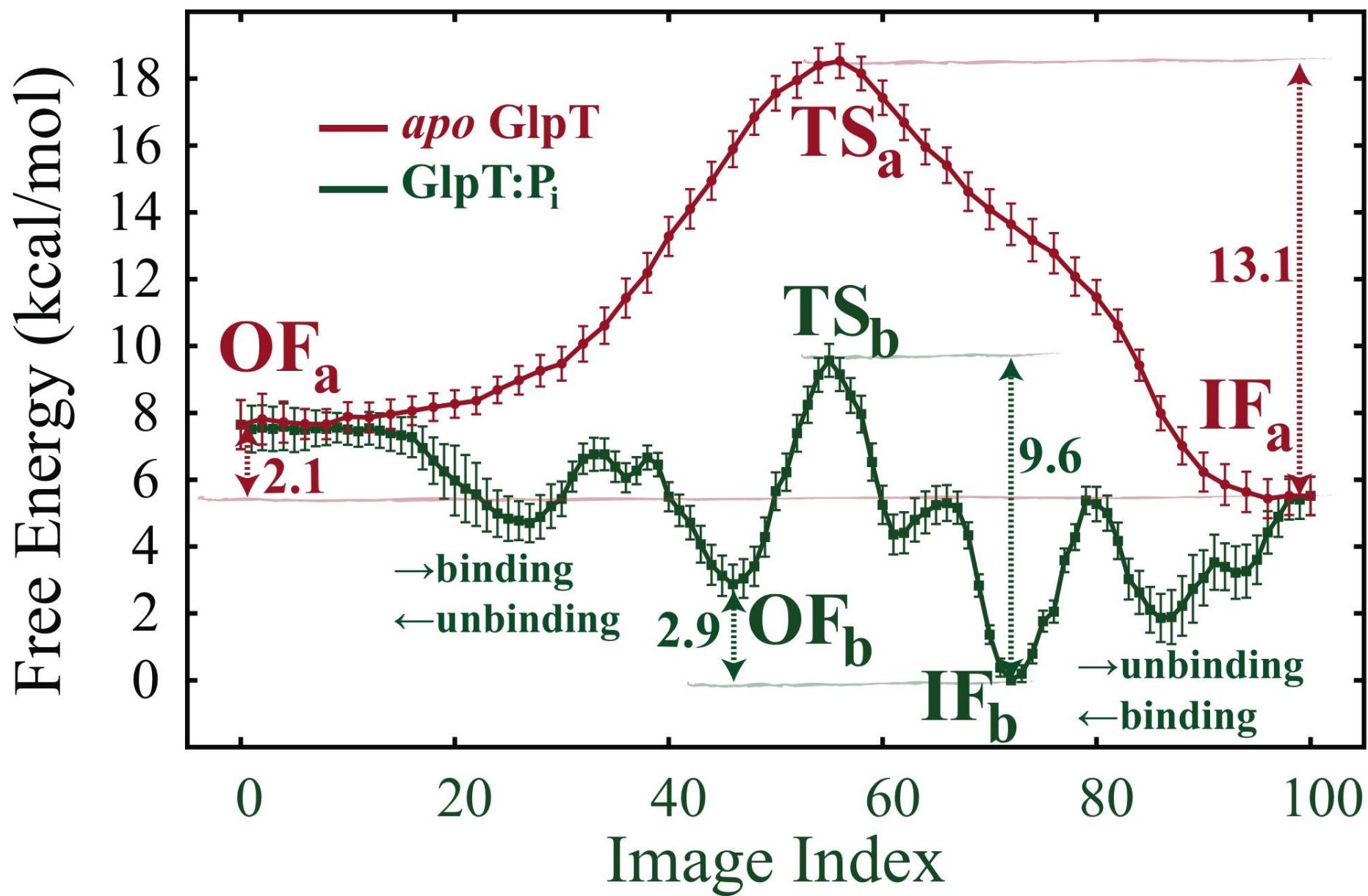
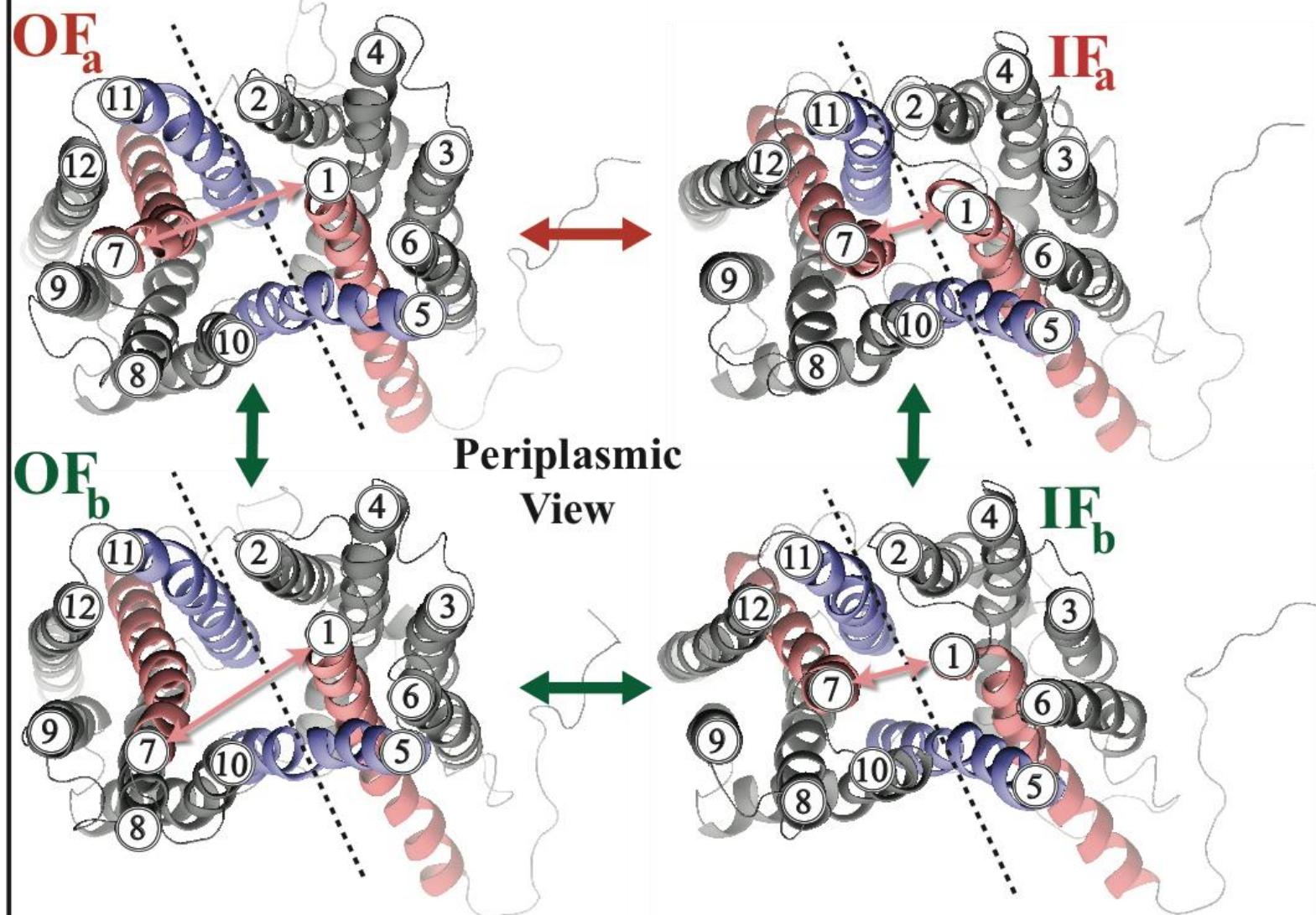
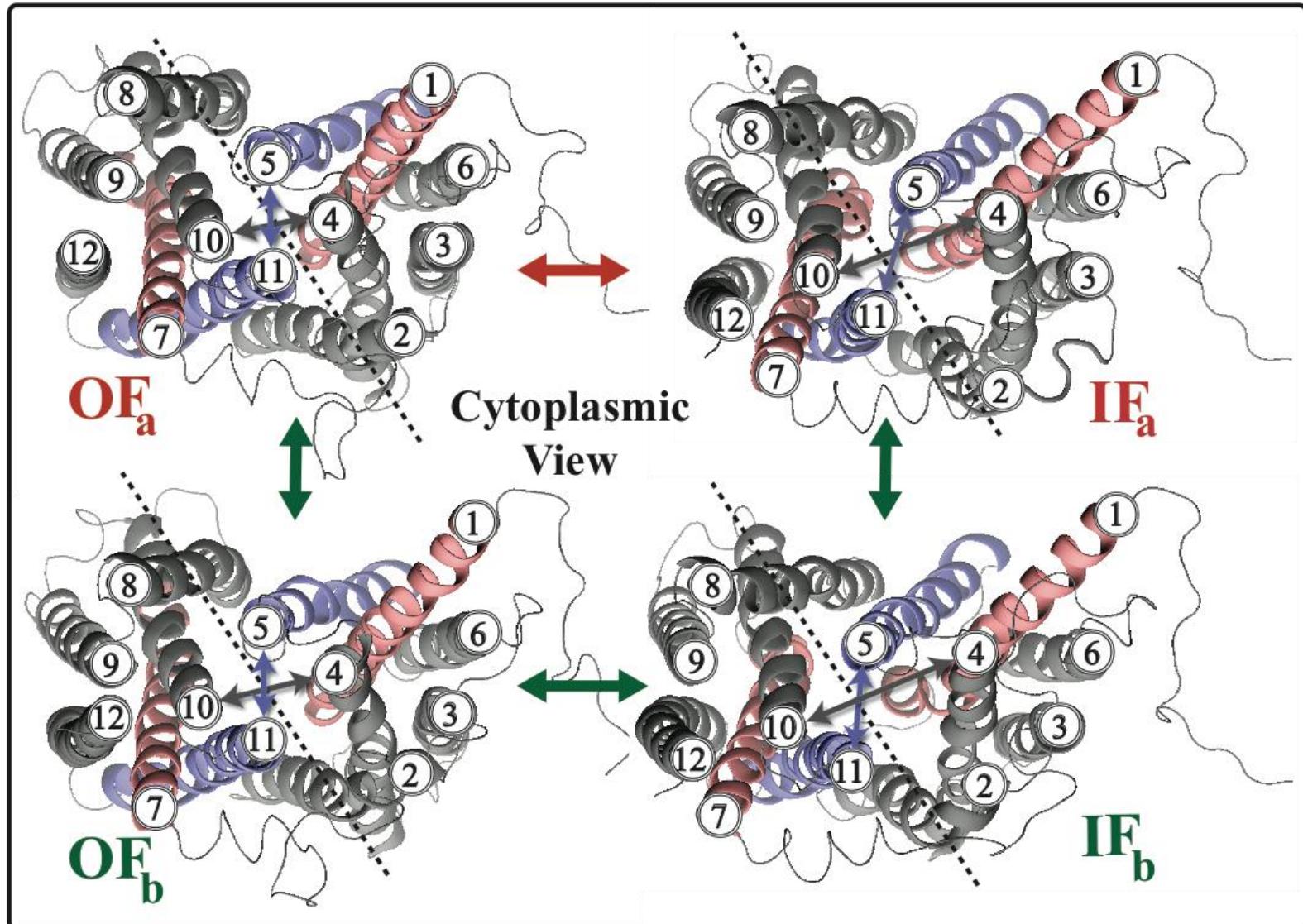


Image Index

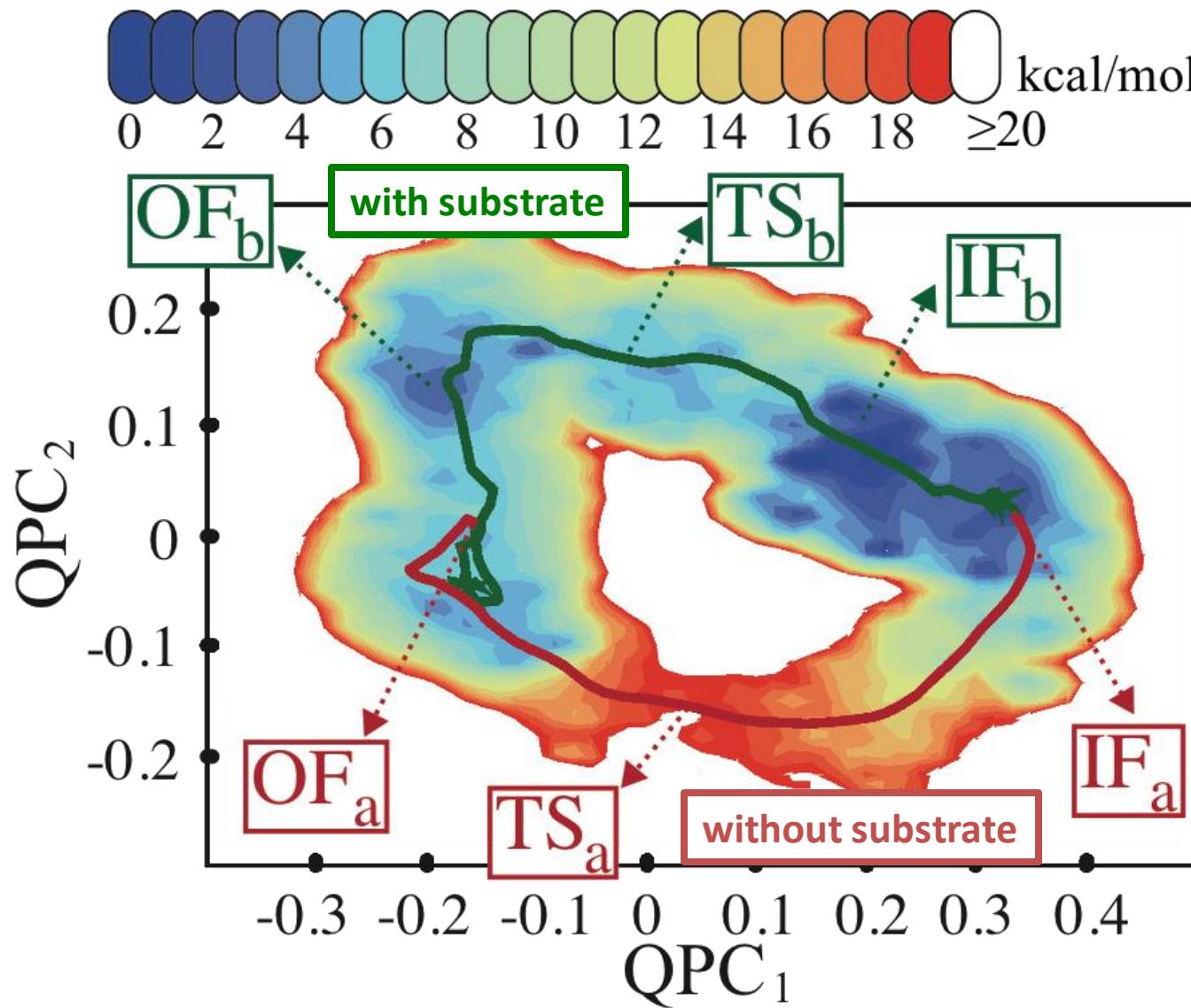






Distinct conformational transition pathways

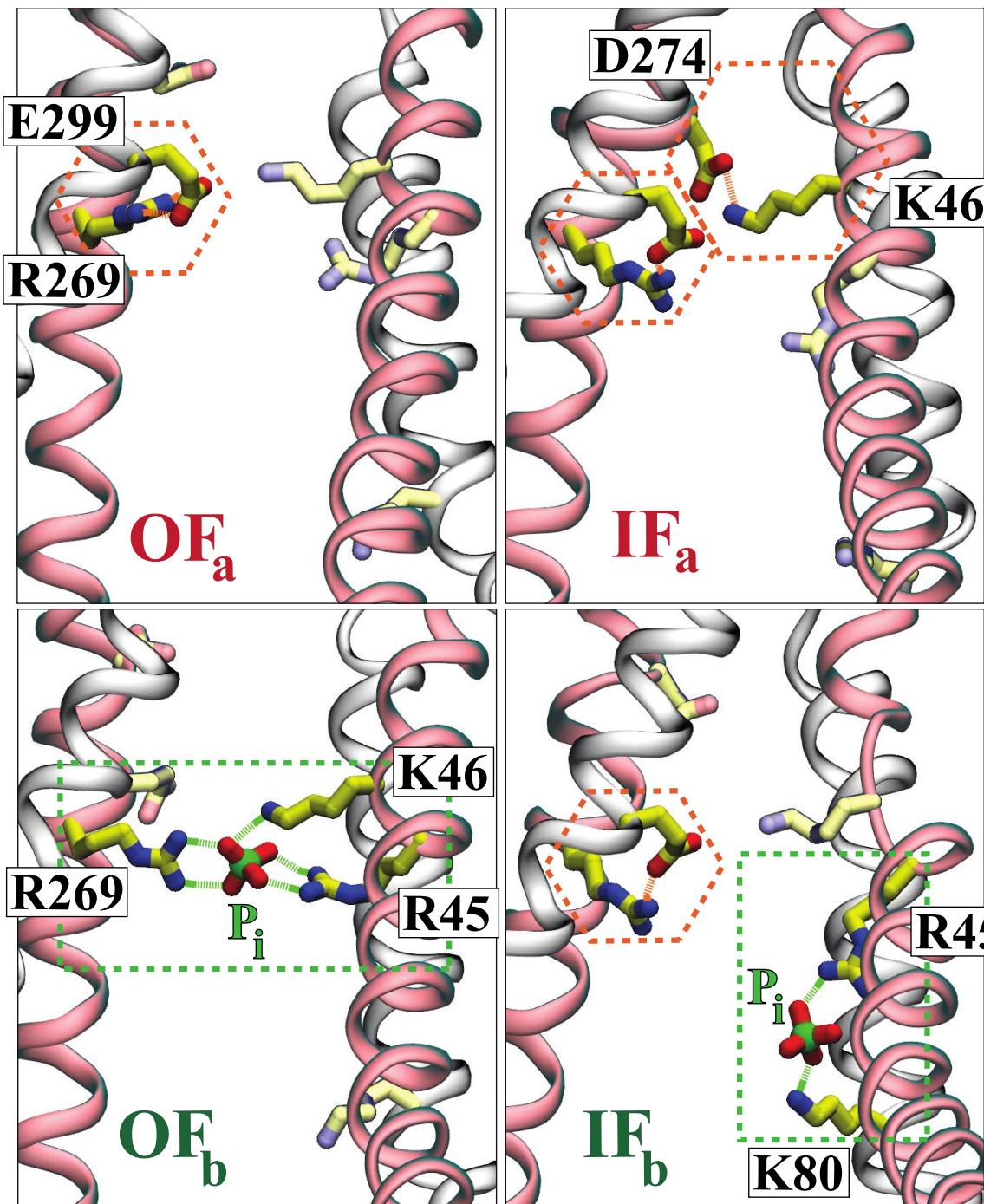
Quaternion-based principal components (QPCs) represent different modes of concerted motions of transmembrane helices.

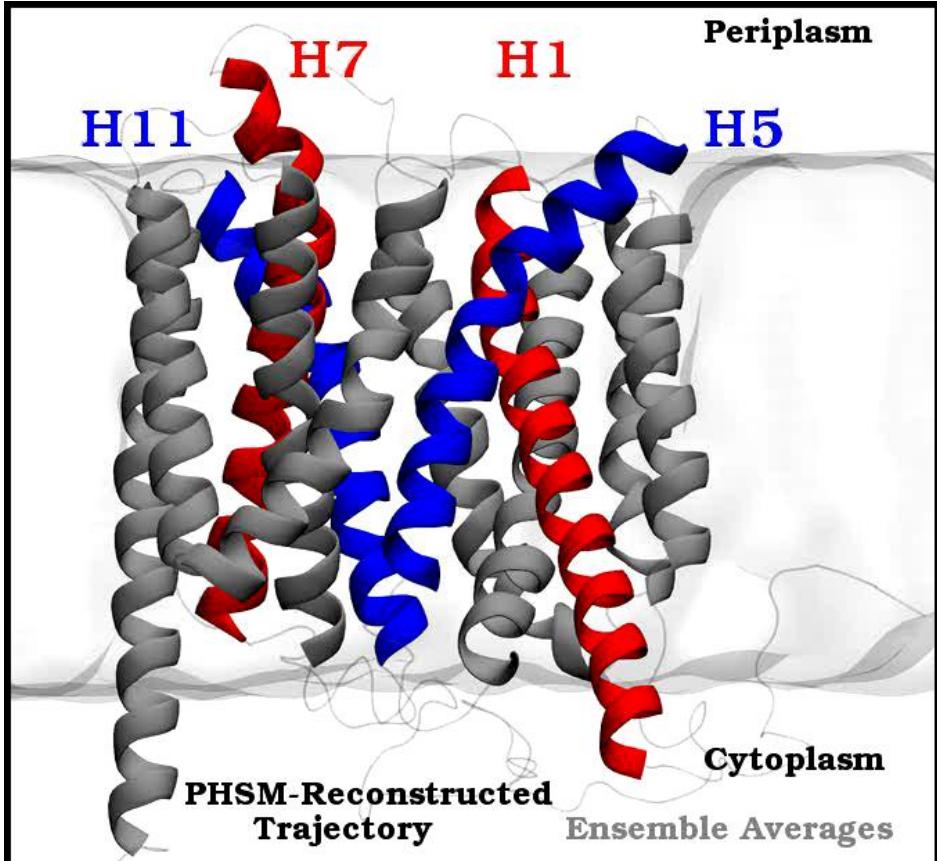


Characterizing protein local conformational changes within the lumen:

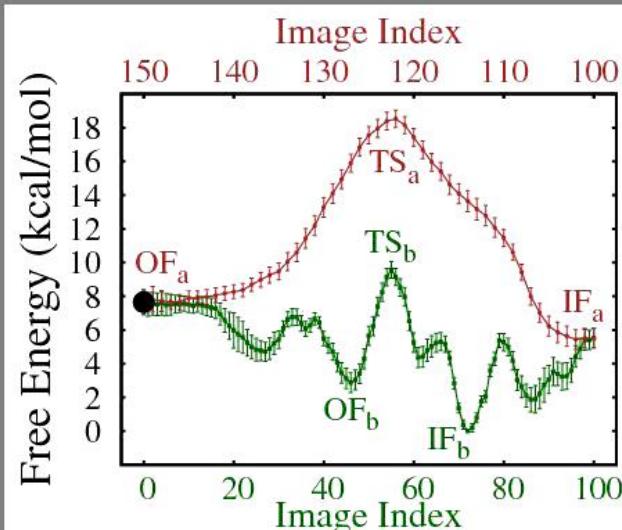
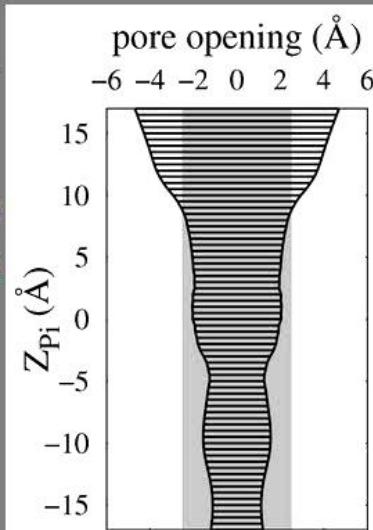
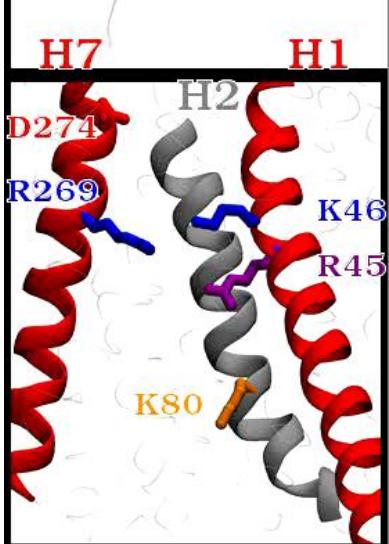
- Salt bridges stabilizing different conformations.
- Residues involved in binding.

Conformational dynamics of the binding site





**Outward-Facing
*apo***



Summary

- **Reconstructed thermodynamic cycle of GlpT**
 - Alternating access mechanism characterized (atomic level)
 - Substrate binding lowers the IF-OF transition barrier
 - Substrate binding changes the IF-OF transition pathway
 - Coupling between local and global conformational changes
- **Reconstructing transport cycles in membrane transporters using enhanced sampling techniques and petascale computing**

Acknowledgement

Emad Tajkhorshid

Giray Enkavi

Tajkhorshid Lab, Beckman Institute, UIUC



Extreme Science and Engineering
Discovery Environment

Rocker-switch model

